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Docket No. D.P.U. 91-131

Investigation by the Department of)
Public Utilities on its own motion)
as to the environmental externality)
values to be used in resource cost-)
effectiveness tests by electric)
companies subject to the)
Department's jurisdiction.)

COMMONWEALTH OF MASSACHUSETTS
DEPARTMENT OF PUBLIC UTILITIES

JOINT TESTIMONY OF

PAUL CHERNICK
and
EMILY CAVERHILL

Resource Insight, Inc.

ON BEHALF OF
BOSTON GAS COMPANY

October 4, 1991

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Attachments

1. Resume of Paul L. Chernick
2. Resume of Emily J. Caverhill
3. Evaluation of New York State Energy Plan Cost Analysis
4. Fact Sheet on TEMIS externalities model

1. WITNESS IDENTIFICATION AND QUALIFICATIONS

Q: Mr. Chernick, please state your name, occupation and business address.

A: I am Paul L. Chernick. I am President of Resource Insight, Inc., 18 Tremont Street, Suite 1000, Boston, Massachusetts. Resource Insight, Inc. was formed in August 1990 as the combination of my previous firm, PLC, Inc., with Komanoff Energy Associates.

Q: Summarize your professional education and experience.

A: I received a S.B. degree from the Massachusetts Institute of Technology in June, 1974 from the Civil Engineering Department, and a S.M. degree from the Massachusetts Institute of Technology in February, 1978 in Technology and Policy. I have been elected to membership in the civil engineering honorary society Chi Epsilon, and the engineering honor society Tau Beta Pi, and to associate membership in the research honorary society Sigma Xi.

I was a Utility Analyst for the Massachusetts Attorney General for over three years, and was involved in numerous aspects of utility rate design, costing, load forecasting, and the evaluation of power supply options. Since 1981, I have been a consultant in utility regulation and planning, first as a Research Associate at Analysis and Inference, after 1986 as President of PLC, Inc., and in my current position at Resource Insight., I have advised a variety of clients on utility matters. My work has considered, among other things,

1 the need for, cost of, and cost-effectiveness of prospective
2 new generation plants and transmission lines; retrospective
3 review of generation planning decisions; ratemaking for plant
4 under construction; ratemaking for excess and/or uneconomical
5 plant entering service; conservation program design; cost
6 recovery for utility efficiency programs; and the valuation
7 of environmental externalities from energy production and use.
8 My resume is Attachment 1 to this testimony.

9 Q: Have you testified previously in utility proceedings?

10 A: Yes. I have testified approximately eighty times on utility
11 issues before various regulatory, legislative, and judicial
12 bodies, including the Massachusetts Department of Public
13 Utilities, the Massachusetts Energy Facilities Siting Council,
14 the Vermont Public Service Board, the Texas Public Utilities
15 Commission, the New Mexico Public Service Commission, the
16 District of Columbia Public Service Commission, the New
17 Hampshire Public Utilities Commission, the Connecticut
18 Department of Public Utility Control, the Michigan Public
19 Service Commission, the Maine Public Utilities Commission, the
20 Minnesota Public Utilities Commission, the South Carolina
21 Public Service Commission, the Federal Energy Regulatory
22 Commission, and the Atomic Safety and Licensing Board of the
23 U.S. Nuclear Regulatory Commission. A detailed list of my
24 previous testimony is contained in my resume.

25 Q: Have you testified previously on externalities?

1 A: Yes. I have testified extensively on externalities valuation
2 in Massachusetts for the past two and a half years on behalf
3 of the Boston Gas Company. My testimony in Vermont Public
4 Service Board Dockets 5270 and 5330 also included
5 externalities. Additionally, I have testified or prepared
6 comment on externalities valuation and incorporation in
7 California, Ontario, Illinois, Maryland, and Indiana, and have
8 worked on the Conservation Law Foundation/New England Electric
9 externalities collaborative.

10 Q: Have you authored any publications on externalities?

11 A: Yes. I have authored about a dozen publications, listed on
12 my resume, on externalities valuation. I have presented
13 several of these papers at national conference and was invited
14 to the World Clean Energy Conference in Geneva to speak on
15 externalities.

16 Q: Ms. Caverhill, please state your name and business address.

17 A: My name is Emily J. Caverhill, and I am a Research Associate
18 with Resource Insight, Inc., 18 Tremont Street, Suite 1000,
19 Boston, Massachusetts, 02130.

20 Q: Please summarize your professional education and experience.

21 A: I received a Bachelor of Science degree in Chemical
22 Engineering in 1984 from Queens University at Kingston,
23 Ontario. I worked for 2 1/2 years at Petro-Canada Inc. as a
24 Petroleum Engineer in Calgary, Alberta and became a
25 professional member of the Association of Professional
26 Engineers, Geologists, and Geophysicists of Alberta in 1986,

1 of which I am currently a member in good standing. I received
2 a Masters of Business Administration in May, 1989, also from
3 Queens University.

4 I joined Resource Insight (formerly PLC, Inc.) in July,
5 1989. Since then, my primary responsibility has been the
6 valuation of environmental externalities of power generation,
7 with some related work in other aspects of least-cost
8 planning. My work has concentrated on valuing externalities
9 which may affect near-term decisions in utility supply
10 planning, and in solving issues related to the practical
11 application of externalities to all aspects of integrated
12 resource planning.

13 Q: Have you testified previously on externalities?

14 A: Yes. I have testified on externalities valuation in
15 Delaware's Regulation Docket 29 regarding integrated resource
16 planning on behalf of the Public Service Commission Staff,
17 and in a recent Illinois Statewide proceeding on integrated
18 resource planning on behalf of the City of Chicago. I co-
19 sponsored testimony and attended a workshop on behalf of the
20 Coalition for Energy Efficient and Renewable Technologies
21 (CEERT) before the Public Utility Commission in California.
22 I have prepared reports and testimony for numerous other
23 jurisdictions, including Massachusetts, Vermont, Maryland,
24 Indiana, Pennsylvania, and Ontario.

25 Q: Have you authored any papers or publications on the valuation
26 of environmental externalities?

1 A: Yes. I have co-authored several papers on evaluating and
2 monetizing environmental externalities. I have spoken on the
3 evaluation of environmental externalities at several
4 conferences in the United States and Canada, and participated
5 in an international workshop on evaluating externalities in
6 Germany in October, 1990. I am a contributing author to three
7 books on the topic. My papers and publications are listed on
8 my resume.

1 2. INTRODUCTION

2 2.1 Purpose and Scope of Testimony

3 Q: What is the purpose and scope of your testimony?

4 A: The purpose of this testimony is to respond to Massachusetts
5 Department of Public Utilities (DPU) Docket 91-131, which
6 invited interested parties to update, and add to, the
7 externality values adopted by the Commission in DPU Docket
8 89-239. We provide analyses that suggest that some of the
9 externality values chosen by the DPU may be low estimates of
10 the value of reducing those externalities, but do not
11 recommend any changes at this time. We provide some
12 additional emissions factors for gas-fired engines; update
13 national and international efforts on developing full-fuel
14 cycle externalities; recommend an externality value for CFCs;
15 and suggest preliminary externality values for heavy metals
16 and thermal pollution. Finally, we briefly review the
17 activities of several states that have adopted explicit
18 valuation of externalities following the Massachusetts Order
19 in 89-239.

20 Q: How is the rest of your testimony organized?

21 A: Section 3 contains updates to our previous analyses of
22 externalities including updated information on the externality
23 values for SO₂, CO₂, NO_x, particulates and CO. Section 4
24 provides suggested additions to the DPU's list of
25 externalities, including CFCs, heavy metals and cooling water.
26 Section 5 updates the regulatory actions in states that have

1 adopted or otherwise supported explicit valuation of
2 externalities.

3
4 2.2 Background

5 Q: Please describe the background of this proceeding.

6 A: In Massachusetts, the valuation of environmental externalities
7 started with a series of orders (e.g., DPU 86-36-G, DPU 86-67,
8 Phase II) in which the DPU instructed various utilities to
9 incorporate externalities in screening resources, especially
10 DSM. The only utility to respond substantively to this
11 instruction was Boston Gas Company, which filed the results
12 of our first analyses in DPU 88-67, Phase II, early in 1989.

13 Boston Gas also filed our updated externality analyses
14 in DPU 89-239. The Energy Office filed analyses by the Tellus
15 Institute, which adopted our methodology and some of our
16 values. The Department adopted our approach to the valuation
17 of environmental externalities, and adopted specific values
18 for air emissions based on the combined Resource
19 Insight/Tellus analyses.

20 DPU 89-239 directed the electric utilities to consider
21 other environmental effects of power plant operation and
22 downstream operations (e.g., disposal of solid and liquid
23 wastes) (p. 84), invited them to propose values for other
24 environmental externalities (p. 89), and encouraged them to
25 address economic and social externalities (particularly oil
26 imports) on a case-by-case basis and in IRM (pp. 81-82). To

1 date, no electric utility has provided estimates of the values
2 of any other externalities, or the data necessary to estimate
3 such values either from the marginal-cost-of-control approach
4 or from the direct costing approach.

5 Q: What has happened with the valuation of externalities in other
6 jurisdictions since DPU 89-239?

7 A: The Massachusetts DPU has emerged as a leader in a growing
8 national trend to monetize externalities and include them in
9 utility planning. Prior to DPU 89-239, the only regulatory
10 commission using monetized externalities was New York, which
11 had taken a very tentative and cautious first step in this
12 direction. Following DPU 89-239,

- 13 • the California Energy Commission adopted monetized cost-
14 of-control values for various pollutants, for utility
15 planning purposes;
- 16 • the California PUC extended and strengthened the Energy
17 Commission approach, and applied the values to
18 acquisition decisions;
- 19 • the Nevada PSC adopted externality values very similar
20 to those of Massachusetts;
- 21 • the Bonneville Power Administration adopted monetized
22 direct-cost values for several pollutants, for use in
23 resource acquisition;
- 24 • the New Jersey Board of Regulatory Commissioners decided
25 to monetize externalities, and adopted interim values per
26 kWh for screening electric and gas conservation programs;

- the South Carolina PSC required utilities to monetize externalities where possible, and to include non-monetized externalities, in their Integrated Resource Management cost analyses; and
- the staff of the Wisconsin PSC concluded that the PSC's current policy of valuing externalities at 17.6% of direct avoided costs was inadequate, and described monetization as "a more sophisticated approach" recently developed in, among other places, Massachusetts.¹

Thus, the DPU's leadership role in monetization of externalities is helping to move the national state of practice. What the DPU does with externalities will certainly affect emissions in and around New England, as a result of the resource decisions made by Massachusetts utilities. The DPU's leadership also appears to be affecting the actions of other states, and hence emission in those states, some of which are upwind of Massachusetts.

2.3 Summary of Conclusions

Q: Please summarize your conclusions.

A: With respect to the major air pollutants the DPU has previously valued (SO₂, CO₂, NO_x, and particulates), we find that the marginal cost of control is either at least as high

¹Wisconsin PSC Staff have expressed an interest in seeing the Resource Insight valuation analysis explored in the Advance Plan 6 proceeding. In fact, Wisconsin Gas is sponsoring our testimony in that case.

1 as the value selected in DPU 89-239, or higher. New data is
2 sparse, except for CO₂, but what is available indicates that
3 some upward movement in the valuation would be justified.
4 Indeed, if the DPU were setting externality values for the
5 first time, the value of each of the major air pollutants
6 might well be increased by anywhere from 30% (for NO_x) up to
7 a few hundred percent (for the other three). In the interests
8 of continuity, and to allow more time to determine how much
9 of an effect the current values will have on utility planning
10 and acquisition, the DPU might impose much smaller increases,
11 if any adjustment is to be made at all.

12 We have also monetized three new sets of externalities:
13 ozone-depleting chemicals, air toxics, and thermal pollution.
14 The values we develop for these externalities should be
15 promptly incorporated in utility planning and acquisition
16 analyses. Doing so would increase the avoided cost of power
17 from existing oil plants by about 1.4¢/kWh (1¢ for air toxics
18 and 0.4¢ for thermal pollution); increase the environmental
19 costs of new clean coal plants by about 0.4¢/kWh and of
20 combined-cycle plants burning oil for 2 months by about
21 0.07¢/kWh (both for air toxics); and result in more careful
22 treatment of chlorofluorocarbons in DSM programs.

23 We continue to support the incorporation of economic
24 externalities, specifically the oil import premium, in the
25 DPU's externality scheme.

1 3. UPDATES

2 3.1 Sulfur Dioxide

3 Q: Have you updated your valuation of SO₂?

4 A: Yes. Our review of the valuation of utility SO₂ emissions
5 covered four aspects, including the effects of the acid rain
6 provisions of the Clean Air Act Amendments of 1990 (CAAA), the
7 cost of SO₂ emissions to be internalized under the CAAA, the
8 interactions between the internalized SO₂ costs and
9 externalities, and new estimates of the external costs of SO₂
10 emissions.

11 Q: How will the acid rain provisions of the CAAA affect emissions
12 of SO₂ by utilities, and the costs of those emissions?

13 A: The primary provisions, which determine the emissions
14 reductions required, specific control requirements for many
15 existing sources, compliance dates, and outline the principles
16 behind the SO₂ allowance trading system are contained in Title
17 IV, which deals with reducing emissions of acid gases SO₂ and
18 NO_x.

19 The primary purpose of this title is to reduce the
20 adverse effects of acid deposition through a reduction in
21 annual SO₂ emissions of 10 million tons and a reduction in
22 annual NO_x emissions of 2 million tons in the lower 48 states
23 and the District of Columbia.² The legislation obtains SO₂
24 emissions reductions through a combination of retrofit control

25 ²U.S. EPA, Clean Air Act Amendments of 1990 Detailed Summary
26 of Titles. November 30, 1990.

1 requirements on some existing utility plants, and a permanent
2 cap on total national SO₂ emissions through a market based
3 system of emissions allowances, to which all "affected units"
4 (essentially all utility sources larger than 25 MW) of SO₂
5 will be subject.³ An allowance is defined as the
6 "authorization, allocated to an affected unit, to emit, during
7 or after a specified calendar year, one ton of SO₂."⁴ SO₂
8 emission allowances will be allocated to existing units under
9 the Phase I provisions; new units which commence operation
10 after 12/31/95 will be required to obtain allowances from
11 existing units, through allowance trading, or from the EPA
12 Administrator, from whom a limited number of allowances will
13 be available on an annual basis at a fixed price of \$1,500/ton
14 (1990\$). The penalty for non-compliance is set at \$2,000 per
15 ton emissions in excess of the allowances held by the affected
16 source, and the excess emissions will be required to be offset
17 the following year. Beginning in the year 2000, the total
18 number of allowances issued by the EPA is essentially not to
19 exceed 8.9 million tons. If this emissions cap is exceeded,
20 the Administrator can reduce the available allowances on a
21 pro-rata basis for all sources to bring total national
22 emissions below the cap. The intent of Congress was to cap
23 national emissions at 8.9 million tons of SO₂ per annum.

24 ³Industrial sources may also opt-in to the allowance program,
25 and may do so depending on their costs of SO₂ reductions and the
26 market price of allowances.

27 ⁴U.S. EPA (1990).

1 Within several years, the market clearing price of SO₂
2 allowances, where one allowance equals one ton of SO₂
3 emissions for one year, will be established through a market
4 trading mechanism, perhaps including trading on the Chicago
5 Board of Trade. SO₂ allowance prices will be determined by
6 the demand for allowances and the supply of allowances, both
7 of which will be determined by, among other factors:

- 8 • the marginal costs of various methods of SO₂ control
9 (price differentials between high-sulfur and low-sulfur
10 fuel coal and oil, and between coal or oil and gas; costs
11 of scrubbers and other desulfurization technologies,
12 etc.), which determine the willingness of allowance
13 suppliers to supply more allowances, and the willingness
14 of allowance buyers to avoid their own compliance
15 actions;
- 16 • the extent of utility risk aversion and resultant
17 allowance stockpiling;
- 18 • demand for electric energy; and
- 19 • cost differentials between low-sulfur and high-sulfur
20 options for new generation.

21
22 Q: What estimates are available of the cost of SO₂ allowances
23 under Title IV of the CAAA?

24 A: The cost of each allowance will depend on the demand for
25 allowances, which is a function of new coal- and oil-fired
26 power plant construction, retirements and repowerings, and

1 usage of existing units, and on the supply of allowances,
2 which is a function of the cost of low-sulfur fuels and of
3 emission control technologies. These costs may include, for
4 various utilities, the purchase of allowances, the
5 construction of scrubbers (which also add to variable O&M,
6 increase heat rates, and decrease baseload plant capacity),
7 the use of more expensive low-sulfur fuel, conservation, and
8 the substitution of gas for coal and oil.

9 ICF (1989) estimated that allowances would trade for
10 \$651-711/ton SO₂ in 2000, \$527-650 in 2005, and \$575-800 in
11 2010, all in 1988 dollars.⁵ National Acid Precipitation
12 Assessment Program (NAPAP) projects a cost of allowances of
13 about \$800-\$1,200/ton SO₂ (1990\$).⁶

14 The Illinois DENR estimated the cost of several control
15 options to reduce SO₂ emissions in Illinois.⁷ For scrubbing
16 high sulfur coal the costs were in the range \$582-\$1,955/ton
17 SO₂, and for fluidized bed technology the cost was \$755-
18 \$1,397/ ton SO₂.⁸ The Allegheny Power System projects control
19 costs on its system of about \$576/ton SO₂ to meet its Phase I

20 ⁵ICF Resources Inc., Economic Analysis of Title V (Acid Rain
21 Provisions) of the Administrations's Proposed Clean Air Act
22 Amendments (H.R.3030/S.1490), Prepared for the U.S. EPA, September,
23 1989.

24 ⁶NAPAP Key Results, Statement of James R. Mahoney, National
25 Academy of Sciences, September 5, 1990.

26 ⁷Baker D. and Bishop, J., "Analysis of Acid Rain Control
27 Alternatives," Illinois Department of Natural Resources, October,
28 1987.

29 ⁸Figures are in the study year's dollars.

1 requirements and \$782-\$960/ton SO₂ to meets its Phase II.
2 requirements.⁹ Other utilities report intentions to install
3 similar measures. The Keystone-Conemaugh Owners Committee
4 (KCOC) is considering scrubbing two Conemaugh units to comply
5 with Phase I emissions requirements at an average cost of
6 about \$500/ton SO₂.¹⁰ Baltimore Gas and Electric, a member of
7 the KCOC, is considering fuel switching to 0.8% sulfur coal
8 at Crane, at a cost of about \$540/ton SO₂ to meet its Phase I
9 requirements.¹¹ BGE estimates that purchases of allowances
10 from other Conemaugh parties would be an alternative to fuel
11 switching at Crane, at a cost of \$497-\$1,874/ton SO₂.¹² A
12 preliminary analysis done by Florida Power Corporation showed
13 that switching to lower sulfur coals at its Bartow Plant oil
14 plant and the coal-fired Crystal River Units 1 and 2 should
15 be sufficient to meet its Phase II emissions limits without
16 scrubbing or purchasing allowances. Finally, under the CAAA,

17 ⁹Figures expressed in 1990\$. Allegheny Power System, "West
18 Penn Power Company's strategy to comply with the requirements of
19 the Clean Air Act Amendments of 1990," February, 1991. APS
20 expected to meet its Phase I targets through wet FGD on its
21 Harrison plant and its Phase II targets through fuel switching
22 and/or scrubbers on its Hatfield units.

23 ¹⁰Information is from a letter to Gregory Carmean of the
24 Maryland Public Service Commission regarding Clean Air Act Title
25 IV Compliance, April 12, 1991.

26 ¹¹Ibid.

27 ¹²Ibid. Under the Phase I provisions for substitutions, a
28 utility can reallocate its required emissions reductions to another
29 unit "under the control of (the same) owner operator." Therefore,
30 BGE claims purchases from Conemaugh to be a reduction option for
31 Crane.

1 the Administrator will set up a "Direct Sale Subaccount"
2 within the "Special Allowance Reserve," which will contain
3 50,000 allowances annually for sale at \$1,500 per allowance
4 in 1990 dollars.¹³

5 Q: How do the SO₂ costs internalized through the CAAA allowance
6 requirements interact with the costs of the external
7 environmental effects of SO₂ emissions?

8 A: Sulfur dioxide (SO₂) is a special case for externality
9 valuation because of the effects of the impending allowance
10 market. Through this emissions-trading mechanism some of the
11 costs of SO₂ emissions will be internalized.

12 Some argue that once the allowance trading mechanism is
13 in place, utilities will be paying for the sulfur they emit,
14 and the external cost of SO₂ emissions will drop to zero.¹⁴
15 This assertion is generally made independent of any estimate
16 of the market value of emissions: in this view, the
17 environmental cost will be whatever the market-clearing price
18 for allowances turns out to be. Proponents of this position
19 may advance two arguments in its support:

- 20 • First, it can be argued that the market price of
21 allowances reflects society's willingness-to-pay for
22 emissions reductions. If Congress had wanted higher
23 levels of controls, it would have ordered greater

24 ¹³Clean Air Act Amendments of 1990, Title IV, Section 416(c).
25 This subsection requires the price of allowances to rise with
26 inflation based on the Consumers Price Index.

27 ¹⁴Goldsmith, M.W., Testimony in ICC Docket No. 91-0050.

1 reductions in emissions, resulting in a tighter allowance
2 market and a higher allowance price.

- 3 • Second, under Title IV of the CAAA, reductions in sulfur
4 emissions by a Massachusetts utility will not result in
5 reduced national emissions. The allowance freed up by
6 the reduced emission will simply be resold to some other
7 utility or QF, increasing emissions elsewhere. Hence,
8 there is no net environmental benefit from SO₂ reductions
9 due to the resource decisions of Massachusetts utilities,
10 only a transfer of pollution from Massachusetts to
11 another region.

12 Q: Is this a valid position?

13 A: No. This argument would only hold true if the cost of buying
14 an allowance fully internalized the cost of SO₂ emissions.
15 The allowance system is only designed to reflect a portion of
16 the costs of SO₂ emissions. Once a working market emerges for
17 emissions allowances, the market value of emissions will be
18 an estimate of society's willingness to pay for emissions
19 reductions at the national level, for acid rain control. This
20 is a minimum national level, and ignores both regional
21 differences and all considerations other than acid rain.
22 Sulfur emissions have health and visibility effects that vary
23 in their importance across the country. Emissions in the
24 populous Northeast are likely to be more costly to society
25 than emissions in the Great Plains, so there is apt to be a

1 social saving from transferring a pound of SO₂ from
2 Massachusetts to Nebraska (for example).

3 Indeed, the Bonneville Power Administration (BPA) has
4 found just such a gradient within its own extensive service
5 territory. In the sparsely populated eastern portion of the
6 service territory, from Montana through eastern Washington and
7 Oregon, BPA estimates a social cost of SO₂ emissions only
8 about 10% of the value it estimates for emissions in the most
9 densely settled portion of the service territory, western
10 Washington.¹⁵

11 From the beginning of the externality valuation process,
12 in DPU 89-239, the DPU has recognized that, while the Federal
13 government may set some national requirements, such as the
14 New Source Performance Standards (NSPS), Massachusetts may
15 place a higher value on a pollutant and impose more stringent
16 conditions. Emissions of NO_x, VOCs, and PM were valued in
17 DPU 89-239 at the cost of controls required in Massachusetts,
18 but not nationally. At the time, no Massachusetts-specific
19 SO₂ control requirements had been priced out, but there is no
20 reason that the same approach applied to these other
21 pollutants cannot be applied to SO₂.

22 The allowance requirements under the CAAA will not always
23 be the most demanding or expensive Federal SO₂ requirements.
24 In many cases, Federal Prevention of Serious Deterioration

25 ¹⁵Bonneville Power Administration, Environmental Costs and
26 Benefits. February 22, 1991.

1 (PSD) and NSPS regulations may require emission controls
2 independent of the allowance system. Indeed, EPA has required
3 scrubbers at the Navaho power plant, to improve visibility in
4 the Grand Canyon. The Navaho plant burns very low sulfur
5 coal, and would probably not be scrubbed under the allowance
6 trading system; the owners would find it cheaper to buy
7 allowances.¹⁶

8 States are also free to impose environmental requirements
9 in addition to those of the Federal government. Just as
10 individual states adopt other pollution control regulations
11 that are stricter (and more expensive) than the minimum
12 federal standards, the states may initiate independent SO₂
13 requirements that imply a value of reducing SO₂ emissions that
14 is much higher than the federally defined regulations. For
15 example, the NYSEP expects that state legislation will require
16 a reduction of 100,000 tons beyond the state's Clean Air Act
17 allowance level of 380,000 tons.

18 Utilities in all states will have to bear at least the
19 cost of the federal SO₂ allowances. In some states, including
20 Massachusetts, the federal SO₂ emissions trading system will
21 not be the limiting factor in utility SO₂ emissions, specific
22 state- and other federal-level regulations will supersede
23 them.

24 ¹⁶According to The Energy Report, September 30, 1991, p. 693,
25 the scrubbers will cost \$89 million per year for 63,000 tons per
26 year of reduction, or \$1,400/ton.

1 Each of these other regulations will direct and channel
2 the trading of emission allowances under Title IV of the CAAA.
3 SO₂ emissions will tend to be reduced more in areas with
4 strict regulations and less in areas with looser regulations.

5 Q: What new estimates are available of the external costs of SO₂
6 emissions?

7 A: We have three new estimates to add to the analysis. First,
8 we can check the valuation of SO₂ by comparing it to the DPU's
9 valuation of particulate matter (PM). Table 3.1.1 shows the
10 ambient air quality standard (AAQS) that can be stated
11 consistently for particulate matter (PM) and for SO₂. The PM
12 standard is the maximum 24-hour arithmetic average
13 concentration of particulates that is acceptable under the
14 Massachusetts AAQS.¹⁷ The SO₂ value is the national AAQS,
15 which is accepted by Massachusetts. The PM standard is
16 stricter. In order to cause the same concern as 150 g of PM,
17 there must be 365 g of SO₂ in the air. Thus, each pound of
18 SO₂ contributes only 41% as much to exceeding the AAQS as does
19 a pound of PM. Hence, if PM is worth \$2/lb in 1989\$ (as the
20 DPU found in DPU 89-239), SO₂ would appear to be worth
21 \$0.82/lb. This is about 10% higher than the 75¢/lb estimated
22 in DPU 89-239.

23 Second, the Draft New York State Energy Plan (1991)
24 roughly estimates mortality damages to New York citizens after
25 the CAAA Title IV requirements are met at \$2,244 per ton of

26 ¹⁷The national AAQS is the same number, but for PM10.

1 SO2 emitted based on dose-response estimates and New York
2 State average population density.¹⁸ Before the CAAA
3 reductions, the NYSEO would apparently estimate that the
4 effects would be greater. The estimate does not include acid
5 rain effects, morbidity, visibility, or any other effect.
6 However, for comparison purposes, Massachusetts' average
7 population density of about 715/mi² is about twice New York's
8 (365/mi²). Simply scaling the New York analysis to
9 Massachusetts' population density would place damages at
10 around \$4,500/ton SO₂ in Massachusetts after the CAAA cap is
11 in place.

12 Third, we now have a Massachusetts-specific estimate of
13 the costs of requirements to control SO₂ emissions.
14 Massachusetts DEP now effectively requires new gas-fired power
15 plants using #2 oil as a winter-period or backup fuel to use
16 oil containing no more than 0.2% S.¹⁹ Boston Edison attempted
17 to license Edgar using 0.3% S oil, but was essentially
18 required by DEP to use 0.2%. Boston Edison estimated that the
19 cleaner fuel would cost \$1,185/ton SO₂ reduction, and accepted
20 it as cost-effective. Table 3.1.2 shows the annual cost of

21 ¹⁸The Commission expressed a preference for damage cost
22 estimates in its Order in DPU 89-239 (at 83). This initial
23 estimate is admittedly very rough, and New York is embarking on a
24 four year, \$1.3 million project to estimate better damage figures.
25

26 ¹⁹The plants required to use 0.2% include L'Energia, Masspower,
27 Everett Energy, West Lynn, and Bellingham. Ventron has conditional
28 approval to use 0.3%S #2 for up to 2000 hours, and very clean
29 (0.04%S) kerosene for 1000 hours.

1 Edison's estimated 0.55% cost differential for the lower-
2 sulfur oil. The cost of SO₂ reduction rises from slightly
3 over \$1000/ton in 1994 (Edgar's first possible year of
4 operation) to nearly \$1900/ton in 2014 (the end of the
5 proposed contract between Edgar Energy and Boston Edison), all
6 in 1990\$. DEP staff has indicated that the agency is
7 considering requiring still lower SO₂ levels, at higher costs.

8 Table 3.1.3 summarizes our new estimates. All three of
9 these approaches indicate that the \$1500/ton value selected
10 in DPU 89-239 is more likely to be understated than
11 overstated. The health-based valuations from the first two
12 approaches should logically be added to the acid-rain-based
13 valuation of the allowances.

14 Q: What actions do you recommend the DPU take with regard to the
15 valuation of utility SO₂ emissions?

16 A: The DPU should require that all utilities incorporate in their
17 estimates of direct avoided costs (for evaluating all
18 incremental resources) both the costs of complying with Title
19 IV of the CAAA (e.g., higher fuel costs, lower coal plant
20 capacity) and the incremental costs of allowances sold or not
21 purchased due to the resource. Allowance requirements should
22 be computed for emission levels following the other compliance
23 actions included in the analysis. The incremental costs of
24 allowances might be about \$800/Ton (1990\$ or 1991\$) starting
25 in the year 2000.

1 The DPU might reasonably increase its estimate of the
2 total environmental value of SO₂. A cost of \$2,000/ton would
3 be more consistent with the available data. On the other
4 hand, the DPU's current estimate is not unreasonably low.

5 Whatever total valuation the DPU selects, it should be
6 used in the period 1991-1999. Beginning in 2000, once Title
7 IV is fully implemented, the internalized allowance cost
8 should be subtracted from the externality value to reflect the
9 internalized portion of the cost of SO₂ emissions; the
10 difference between the total value and the allowance cost
11 remains as the external portion of the cost.²⁰

13 3.2 Carbon Dioxide

14 Q: What major issues surround the CO₂ valuation debate?

15 A: The major issues are uncertainties in (1) the analysis of
16 anthropogenic greenhouse effects, including:

- 17 - whether global warming has started,
- 18 - the rate of future warming,
- 19 - and the economic and environmental consequences of global
20 warming;

21 and (2) the effectiveness of reducing carbon dioxide emissions to
22 slow the rate of warming and diminish potentially deleterious
23 environmental effects, economic damage, and the costs of
24 controlling additional damage.

25 ²⁰If the allowance cost exceeded the externality value, then
26 the allowance cost (which will be an internalized cost) would
27 replace the externality value.

1 Q: Where does the current debate stand?

2 A: It is widely recognized that large reductions in net
3 anthropogenic CO₂ emissions, especially by the developed
4 countries, will be necessary to slow global warming to
5 tolerable rates, even with the phase-out of CFC's and controls
6 on other greenhouse gases. Krause, Bach, and Koomey (1989)
7 estimate that reductions of 20% from present levels by 2005
8 or 2010 (or roughly 50% from base case), and reductions of 80%
9 from current levels by 2030 are required from industrialized
10 countries to limit global warming to a tolerable rate. Figure
11 I.6.2 from Krause, Bach, and Koomey is attached as Figure 3.2.
12 The Intergovernmental Panel on Climate Change (1990) estimates
13 that more than a 60% reduction from current emissions levels
14 would be needed to stabilize CO₂ emissions at current levels.²¹

15 Acting on this principle, many countries have adopted CO₂
16 emissions reductions targets (summarized in Table 3.2.1) and
17 some have planned implementation strategies. For example, to
18 reduce its CO₂ emissions 25-30% by 2005, Germany will
19 introduce a carbon tax this year.²² Norway's government has
20 proposed stabilization at 1990 emissions levels by the year
21 2000. Japan's goal is stabilization at 1990 levels of CO₂ by
22 2000. New Zealand plans to reduce carbon dioxide emissions
23 20% by 2000, while Australia's target is stabilization at 1988

24 ²¹IPCC results quoted from Global Environmental Change Report,
25 Vol II, No. 11, 8 June 1990.

26 ²²"Ruling German Parties Agree to Introduce Carbon Tax," Global
27 Environmental Change Report, vol. III, no. 2. January 18, 1991.

1 levels by 2000 and 20% reduction by 2005. Other targets are
2 also listed in Table 3.2.1.

3 Q: How was the original \$22/ton CO₂ value adopted by the DPU
4 derived?

5 A: In Chernick and Caverhill (1989) we developed a range of tree-
6 planting costs for the U.S. based on reasonable assumptions
7 about carbon uptake per acre and planting and maintenance
8 costs per acre, of 2-10 cents/lb carbon.²³ From that range of
9 costs, we chose 4.0 cents/lb carbon (\$80/ton carbon), or
10 \$22/ton CO₂ as our estimate for an externality value for CO₂.
11 Subsequent estimates of tree planting costs, when corrected
12 for comparability, confirm the reasonableness of this range.²⁴

13 Q: How have you updated this estimate?

14 A: We have gathered estimates from several studies that compute
15 (or allow us to compute) a marginal cost for reducing carbon
16 dioxide emissions. Some of the studies directly estimate the
17 marginal cost of achieving various reductions in CO₂
18 emissions. For other studies, we have estimated the marginal
19 cost for the cost of the most expensive measure(s) identified
20 as necessary to achieve the desired reduction in CO₂

21 ²³Chernick, P. and Caverhill, E., "The Valuation of
22 Externalities from Energy Production, Delivery, and Use." Report
23 to the Boston Gas Company, December 22, 1989.

24 ²⁴Chernick, P. and Schoenberg, J., "Determining the Marginal
25 Value of Greenhouse Gas Emissions," Energy Developments in the
26 1990s: Challenges Facing Global/Pacific Markets. International
27 Association for Energy Economics; Honolulu: July 1991.
28
29

1 emissions. Among the estimates we consider are those of
2 microeconomic studies, based on estimates from a specific
3 technology or technologies, and those of macroeconomic
4 studies, based on taxation. Unless stated otherwise, all
5 costs are in 1990 dollars, assuming 4% annual inflation.

6 Q: Have you updated your CO₂ cost analysis using this additional
7 information?

8 A: Yes, we have updated our analysis to include additional
9 international CO₂ targets and costs. We conclude from this
10 analysis that the DPU externality value of \$22/ton CO₂ is
11 probably far below the marginal cost of control measures other
12 countries are implementing to achieve targeted CO₂ reduction
13 targets.

14
15 a. Microeconomic Analyses

16 Q: What is a microeconomic analysis as applied to CO₂ costs?

17 A: Microeconomic analyses derive a value per amount of carbon
18 dioxide reduction from the costs and effectiveness of
19 technological strategies, such as conservation, cogeneration,
20 fuel substitution, renewable energy, and tree-planting. The
21 microeconomic analyses we have reviewed were performed by the
22 World Wildlife Fund (1990), the Danish Ministry of Energy
23 (1990), Manne and Richels (1989), Naill, Belanger, and
24 Petersen (1990), Nordhaus (1991), New York State Energy Office
25 (1989, 1991), and Oregon Department of Energy (1990).

26 Q: What did the World Wildlife Fund study find?

1 A: World Wildlife Fund (1990) compiled studies of the costs of
2 reducing several countries' carbon dioxide emissions by 2005
3 to 80% of 1988 levels. With funding from the U.S. E.P.A.,
4 climate change analysts from industrialized countries
5 undertook studies to calculate reduction costs in their native
6 countries. In most cases, the authors focused on the
7 possibility and costs of attaining a 20% reduction in
8 emissions by 2005. The results appear in Table 3.2.2 [note
9 13], and the calculations in Table 3.2.3.²⁵ The measures
10 reported are those discussed by the various authors. The
11 carbon dioxide reduction costs for the Eastern European
12 countries are much lower than for the U.S. and the U.K.: the
13 result from uncertainties in the exchange rates and/or the low
14 current efficiencies of Eastern European economies. As shown
15 in Table 3.2.2, many measures are necessary to achieve the
16 desired reduction goal.

17 Q: How were the Danish Ministry of Energy estimates of the cost
18 of CO₂ control developed?

19 A: The Danish Ministry of Energy sets up three potential
20 scenarios for future energy and economic development. The
21 first, the supply scenario, concentrates on changing sources
22 and uses of fuels to achieve carbon emissions reductions.
23 The second, the environment scenario, assumes the most

24 ²⁵World Wildlife Fund (1990) also includes chapters on France,
25 Hungary, Canada, and Japan. Japan's estimated cost is mentioned
26 later in this testimony, while the other countries do not appear
27 in the table because of inconsistencies or omissions within the
28 chapters.

1 important consideration in energy supply decisions will be
2 their environmental impacts. The third, the economy scenario,
3 gives priority to the socioeconomic effects of these
4 decisions.

5 Q: How do these three scenarios differ from base case projections
6 of energy capacity and savings in 2000?

7 A: The difference is the exclusion of new coal plants from all
8 three scenarios mentioned previously. The base case scenario
9 assumes 1400 MW of power will be supplied from new coal
10 plants. To compensate, cogeneration replaces coal-fired
11 power, represents a two- to three-fold increase over the
12 amount of cogeneration present in the base case scenario.
13 The supply, energy, and economic scenarios also postulate wind
14 energy's supplying between 1 1/2 to 2 1/2 times the power as
15 in the base case. All three cases offer substantial increases
16 in residential, commercial, and industrial conservation.

17 Q: How did you obtain a marginal cost from the Danish Ministry
18 of Energy report?

19 A: The Danish Ministry of Energy (1990) offers estimates of the
20 average costs of CO₂ emissions reductions as well as average
21 costs of specific measures. To obtain a marginal cost
22 estimate, we examined the differences between the second most
23 expensive scenario - the economy scenario - and the most
24 expensive one - the environmental scenario. As seen in Table
25 3.2.2, and calculated in Table 3.2.4, the incremental costs

1 of reduction - between \$50-\$250/ton CO₂ - are similar to those
2 of the other Western countries summarized in Table 3.2.2.

3 Q: What do Manne and Richels estimate?

4 A: Manne and Richels (1989) estimate the average cost of reducing
5 CO₂ emissions 20% from 1990 values by 2020. The marginal cost
6 would be higher. Their base case is a highly constrained
7 energy scenario, with no new low-cost supply technologies
8 available. The demand side of the scenario assumes no further
9 energy efficiency improvements other than those induced by
10 price changes. In this base case, the average reduction cost
11 peaks at \$86/ton CO₂ in 2020.

12 Q: What do Naill, Belanger, and Petersen estimate?

13 A: Naill, Belanger, and Petersen (1990) estimate potential carbon
14 dioxide emissions reductions from energy efficiency through
15 2030. From conservation supply curves, the authors create
16 categories of "very high" efficiency for the upper end of the
17 curves and "high" efficiency for middle portions of the
18 curves. The incremental costs of reductions are negative for
19 "high" efficiency, and \$69/ton CO₂ for "very high" efficiency.
20 Even the "very high" level would reduce emissions only 28%
21 from the projected levels and would not be sufficient to
22 reduce emissions to 80% of their 1990 levels, even by 2030.

23 Q: What does Nordhaus analyze?

24 A: Nordhaus (1991) derives a marginal cost curve for CO₂
25 reductions by fitting an equation to a range of estimates.
26 Some of the points along the curve are estimates from studies

discussed in this testimony (e.g., Manne and Richels, Jorgenson and Wilcoxon). The curve shows that achieving a 50% reduction in CO₂ emissions from a base case scenario by 2005 (bringing 2005 emissions to 80% of 1990 emissions) would require a tax of approximately \$34/ton CO₂. Along with this curve, Nordhaus uses a marginal cost curve for CO₂ reductions from tree planting and a marginal cost curve for CFC reductions, to create a marginal cost curve for total greenhouse gas reductions.²⁶ Because Nordhaus includes large CFC reductions at low cost, the total greenhouse gas curve is lower than the CO₂ curve derived from the estimates from studies.

Q: What does the New York State Energy Office estimate?

A: The Draft 1991 New York State Energy Plan estimates costs on the order of \$66-\$124/ton CO₂ to achieve goals from stabilization of CO₂ emissions at 1988 levels in 2008 to a 20% reduction in these emissions in the same year. These costs are discussed in Attachment 3.

Q: And the Oregon Department of Energy?

A: An Oregon law passed in 1989 directs several state agencies to develop a strategy to reduce greenhouse gas emissions. A 1990 study by the Oregon Department of Energy (ODOE) presents thirteen technological strategies and the amounts of their contributions toward reducing state carbon dioxide emissions

²⁶Both the CFC curve and the total greenhouse gas curve are expressed in dollars per ton of carbon equivalent.

1 20% below 1988 levels by 2005. Some options examined were
2 increased conservation in all sectors, increased fuel
3 efficiency in cars and light trucks, reduction of single
4 occupancy vehicle trips, vehicular fuel-switching, fuel-
5 switching from electric to gas water heaters, tree planting,
6 and raising participation in recycling programs. ODOE noted
7 uncertainties in mitigation costs, feasibility of reduction
8 measures, and damages caused by climate change, but affirmed
9 its belief that "the threat of global warming is
10 significant."²⁷

11 Table 3.2.2 [note 15] shows the average reduction costs
12 for the strategies for which the ODOE report provided
13 sufficient data to determine costs. It is not clear whether
14 the most expensive of these strategies (at up to \$296/ton CO₂)
15 is needed to meet the state's reduction goals; however, the
16 reduced coal strategy (at about \$50/ton CO₂) is clearly
17 necessary.

18 19 b. Macroeconomic Analyses

20 Q: What is a macroeconomic analysis as applied to CO₂ costs?

21 A: Macroeconomic analyses estimate the cost of reducing carbon
22 dioxide emissions through taxation or similar measures, using
23 computer models of national economies. Carbon taxes have been
24 widely discussed in Europe, the U.S., and Japan. Many of the

25 ²⁷Oregon Department of Energy, Oregon Fourth Biennial Energy
26 Plan: Global Warming Strategy and Two-Year Action Plan, p. 8.
27 October 1990.

1 studies we examine estimate both a tax value, which is the
2 amount required to achieve a certain reduction goal, and a tax
3 cost, which represents the cost to society in terms of lost
4 growth. The value and cost for a single tax are not
5 necessarily equal, and often may be very different from each
6 other. The macroeconomic studies we reviewed are from the
7 World Wildlife Fund (1990), Manne and Richels (1989), Naill,
8 Belanger, and Peterson (1990), and Jorgenson and Wilcoxon
9 (1990).

10 Q: What did the World Wildlife Fund study find?

11 A: World Wildlife Fund (1990) asserts that Japan's existing high
12 efficiency and low energy intensity make further CO₂
13 reductions through technological means difficult. The study
14 estimates that a carbon tax sufficient to maintain 1988
15 emissions levels would reduce gross domestic product growth
16 by .4% per year, or 39 trillion yen (1980 yen) in base case
17 2005 GDP. The marginal cost of reduction is \$6,096/ton CO₂
18 (Table 3.2.2 [note 13]).²⁸

19 Q: What do Manne and Richels estimate?

20 A: Manne and Richels (1989), working within the parameters of
21 the highly constrained energy scenario described earlier,
22 estimate the rates of a carbon tax necessary to achieve a 20%
23 reduction from 1990 carbon emissions levels by 2020. The tax
24 rises very sharply, peaks at \$149/ton CO₂ in 2020, and falls

25 ²⁸The April 22, 1991 exchange rate is 137.65 yen/\$.

1 to \$65/ton CO₂ in 2040. The tax would remain fairly stable
2 for the rest of the century.

3 Q: What do Naill, Belanger, and Petersen estimate?

4 A: Naill, Belanger, and Peterson (1990) consider several taxation
5 levels, as well as a coal-plant efficiency incentive. Under
6 the incentive approach, coal plants not meeting efficiency
7 standards of the best commercially available technology
8 receive a penalty based on the difference in efficiencies.
9 The incremental reduction costs of these approaches appear in
10 Table 3.2.2 [note 2] (calculated in Table 3.2.5). The study
11 explicitly shows that the \$568/ton C tax would achieve the
12 goal of a 20% reduction from 1989 CO₂ emissions by 2020 at a
13 social cost of \$273/ton CO₂. The taxes in the range of \$227-
14 \$364/ton C would reach a 20% reduction around 2030, at costs
15 of \$176-\$219/ton CO₂. The other taxes and the coal incentive
16 are much less effective in achieving emissions reductions.

17 Q: How do Jorgenson and Wilcoxon obtain a cost estimate for a
18 carbon tax?

19 A: Jorgenson and Wilcoxon (1990) estimate the U.S. carbon taxes
20 necessary for achieving three reduction targets, described
21 below. The tax is determined by their model, which accounts
22 for 35 industries, household consumption, investment and
23 capital accumulation, government tax rates and nontax
24 receipts, budget deficit, and foreign trade elasticities.
25 Immediately stabilizing carbon emissions at their 1990 level
26 of 1,576 million tons requires a tax that peaks at \$5/ton CO₂

1 in 2020. The second goal, decreasing carbon emissions to 80%
2 of the 1990 level by 2005, requires a maximum tax of \$15/ton
3 CO₂ in 2020. Finally, taxing emissions beginning in 2000 to
4 maintain that year's base level would cost at most \$3/ton CO₂
5 in 2020.

6 Jorgenson and Wilcoxon estimate the effects of these
7 taxes in terms of percentage change from base case CO₂ and GNP
8 levels. Figure 3.1 combines these percentage reductions with
9 assumptions of 2% annual base case increases in both GNP and
10 carbon emissions to calculate the marginal cost of shifting
11 from the immediate stabilization goal to the goal of the 20%
12 reduction from 1990 emissions levels.²⁹ The marginal cost of
13 control peaks at approximately \$56/ton CO₂.

14 Q: What are the implications of the costs estimates in these
15 studies?

16 A: The implication is that the cost of the measures required to
17 meet reasonable emissions reductions targets are very high,
18 generally higher than the CO₂ externality values adopted by
19 any state utility commission including the \$22/ton adopted by
20 Massachusetts. While we have not attempted to draw an
21 estimate out of these diverse cost estimates, the
22 Massachusetts' externality value is almost certainly too low
23 to reflect marginal CO₂ emissions reduction measures.
24

25 ²⁹Jorgenson and Wilcoxon do not compute absolute dollar costs
26 or emission tonnage.

1 3.3 Nitrogen Oxides (NO_x)

2 Q: Have you updated the DPU's valuation of NO_x?

3 A: Yes. We have updated the regulatory situation and reviewed
4 estimates of the costs of NO_x controls likely to be required
5 under the CAAA.

6 Q: What is the status of air quality regulation of NO_x, and of
7 the role of utilities in NO_x production?

8 A: Utilities are responsible for approximately 25% of total
9 national NO_x emissions.³⁰ NO_x pollution is linked to acid
10 rain, reduced visibility, excess ambient ozone levels and
11 increased incidence of respiratory ailments. Increased levels
12 of respiratory illness have been positively correlated to
13 increases in ambient ozone concentration (Krupnick, et al.,
14 1990). For one or more of these reasons, NO_x emissions
15 control measures are continually increasing in stringency,
16 both at state and federal levels.

17 Q: Which provisions of the CAAA address NO_x control measures
18 applicable to Massachusetts?

19 A: Several provisions in the CAAA address reductions in NO_x
20 emissions, due to its contribution to the serious effects
21 listed above and its variety of stationary and mobile sources.
22 Seven of the eleven Titles of the CAAA address NO_x emissions
23 directly (through specific control requirements) or indirectly
24 (through research and enforcement provisions).

25 ³⁰U.S. Department of Commerce, 1990 Statistical Abstract of the
26 United States.

1 Q: What regulations does Title I of the CAAA address?

2 A: Title I addresses the issue of attaining minimum ambient air
3 quality for several pollutants including ozone, particulate
4 mater (PM), CO, NO_x, SO₂ and lead (Pb). Included in this title
5 are specific reduction targets for these pollutants, including
6 targets for reductions of the ozone precursors, NO_x and VOCs.
7 Under these provisions, most of Massachusetts is classified
8 as in serious non-attainment for ozone.³¹ As such,
9 Massachusetts, and many areas in the Northeast, are subject
10 to several provisions. Among these are:

- 11 1. Enhanced monitoring of ambient ozone, NO_x and VOCs
12 levels.
- 13 2. Attainment of reasonable progress toward national ambient
14 air quality standard (NAAQS) for ozone, which requires
15 VOC emissions reductions "at least 3% of baseline
16 emissions each year," (averaged over three consecutive
17 years beginning six years after the date of enactment of
18 the CAAA) or equivalent NO_x and VOC reductions which
19 would have at least the equivalent effect on ozone level
20 reduction.³²

21 ³¹Curiously, central Massachusetts is currently considered
22 moderate, rather than serious. The implication of this is that
23 this area must come into compliance sooner (within 6 years of
24 enactment) than the serious-designated areas (nine years).
25 Massachusetts has requested a special waiver making the entire
26 state one serious non-attainment region for compliance purposes.

27 ³²Less than 3% per year can be allowed if the state
28 demonstrates to the Administrator that all feasible measures are
29 being taken.

- 1 3. Enhanced vehicle inspection and maintenance for each
- 2 urbanized area (1980 census population of 200,000 or
- 3 more).
- 4 4. Clean-fuel vehicle program (as defined in part C of Title
- 5 II).
- 6 5. Transportation control, which includes the evaluation of
- 7 vehicle mileage, aggregate emissions, congestion levels,
- 8 and other relevant parameters to determine consistency
- 9 with attainment of State Implementation Plan (SIP)
- 10 targets.
- 11 6. Control of interstate air pollution in the Northeast
- 12 transport region, which includes the states of
- 13 Connecticut, Delaware, Maine, Maryland, Massachusetts,
- 14 New Hampshire, New Jersey, New York, Pennsylvania, Rhode
- 15 Island, Vermont, and the CMSA of the District of
- 16 Columbia. States within such a region must submit a new
- 17 or revised SIP including implementation of enhanced
- 18 vehicle inspection and maintenance and RACT (reasonably
- 19 available control technology). Within three years, the
- 20 Administrator must identify control measures which are
- 21 as effective as vehicle refueling measures, and the state
- 22 SIPs must be revised to incorporate such measures within
- 23 one year of the completion of that study.³³ In addition,

24 ³³Chernick and Caverhill (April, 1990) estimated the cost of
25 stage II vapor recovery devices for vehicle refueling at \$1.02/lb
26 VOC, and noted that other estimates for the same program ranged up
27 to double this estimate (API, 1988). OTA (1989) estimated that the
28 most expensive measures required to meet the CAAA ozone provisions

1 sources within this region that emit greater than 50 tons
2 per year of VOCs will be considered major sources, and
3 at a minimum will be subject to the requirements for
4 major sources under the moderate non-attainment regions,
5 which includes VOC offsets of 1.15 to one.³⁴

6 Q: What issues does Title II address?

7 A: Title II reduces mobile source emissions of NMOGs (VOCs), NO_x,
8 particulates and CO. Under the Act, states can choose to
9 adopt either the federal emissions standards, or the more
10 stringent California emissions standards for mobile sources
11 (adopted in California in July 1990).

12 Q: What are the federal and California standards?

13 A: The new federal standards for NMOG, CO and NO_x are defined in
14 Title II of the CAAA, and are shown in Table 3.3.1. The most
15 stringent of the clean fuel requirements for NO_x are for Phase
16 II, light duty trucks and vehicles, which have emissions
17 limits of 0.2 grams per mile for vehicles starting in the
18 model year 2001.

19 The new California standards, referred to as the
20 California Low Emissions Vehicle (LEV) program, set five
21 different emissions limits. A particular vehicle must meet
22 one set of standards, while compliance with a fleet average

23 would be \$6,600/ton (1994\$) or about \$5,500 (1990\$).

24 ³⁴A typical 150 MW gas combined-cycle, a 300 MW coal plant, or
25 a 50 MW oil-fired combined cycle could surpass this threshold.

1 limit will likely mean manufacturers must market vehicles in
2 categories other than only the most lenient.

3 Under the LEV program, vehicles categorized as Standard
4 Vehicles (SV) must meet emissions limits equal to the federal
5 regulations. Emission limits for NMOG, CO, NO_x and
6 formaldehyde for the four other categories of vehicles are
7 also shown in Table 3.3.1.³⁵ Transitional low emission
8 vehicles (TLEV) must limit emissions further than standard
9 vehicles, and emissions limits for low emission vehicles (LEV)
10 are more strict still. Ultra-low emission vehicles (ULEV)
11 must comply with the most strict limits of the program. The
12 California LEV program defines zero emission vehicles (ZEV)
13 as battery-operated electric vehicles. There are no emissions
14 limits since, as the name describes, the zero emission
15 vehicles have zero emissions.³⁶

16 Q: In the LEV program, how are motor vehicles classified into one
17 of the five categories?

18 A: The vehicle's manufacturer chooses under which category it
19 wants a particular model and year vehicle to be certified.
20 The manufacturer must notify the California Air Resources

21 ³⁵NMOG is non-methane organic gases. Information from
22 Massachusetts Department of Environmental Protection, Background
23 Document for Proposed Amendments to 310 CMR 7.00 et seq. September
24 1991.
25
26
27

28 ³⁶That is, the vehicles have no direct emissions other than the
29 incremental emissions from the electric system.

1 Board (CARB) which fuel the vehicle will use. CARB has
2 published a list of the types of fuels and control
3 technologies it expects will be necessary for vehicles to use
4 to belong to anything other than the most lenient category,
5 Standard Vehicle.

6 Q: Have any states in the Northeast adopted or considered
7 adopting the new California standards?

8 A: Yes. Last year, prior to the passage of the CAAA,
9 Massachusetts adopted California emissions limits.³⁷ In March
10 1991, New York also adopted them. Maine and New Jersey are
11 preparing background documents in anticipation of hearings in
12 early 1992. Connecticut has a legislative mandate to study
13 the effects of adopting such a program, while Rhode Island
14 and Vermont have agreed to adopt the program if a majority of
15 the states in the Northeast States for Coordinated Air Use
16 Management district do so.³⁸

17 Q: How does Title IV affect NO_x emissions?

18 A: Title IV deals with reducing total emissions of acid gases SO₂
19 and NO_x, and permanently capping emissions of SO₂. Under this
20 title, low-NO_x burners or the equivalent are required to be
21 retrofit onto existing tangentially-fired and dry bottom wall-
22 fired boilers, and equipment that is at least as cost

23 ³⁷Massachusetts modified the California program slightly by
24 requiring that 2% of a manufacturer's new vehicles sold in the
25 Commonwealth from 1998 onward be zero emission vehicles.

26 ³⁸Information from telephone conversation with A. Marrin,
27 NESCAUM. October 2, 1991.

1 effective, and perhaps more costly, will be required on other
2 types of boilers.³⁹

3 Q: What Massachusetts state regulations govern selection of NO_x
4 control measures?

5 A: Massachusetts uses a top-down approach to determine best
6 available control technology (BACT) for NO_x and other
7 pollutants.⁴⁰ The top-down approach requires the developer of
8 an energy project to use the most effective available
9 technology unless it can establish that measure is not cost-
10 effective, and then it is required to use the next most cost-
11 effective measure. Cost-effectiveness determinations appear
12 to take into account benefits of reductions as well as costs.

13 Q: How do the Massachusetts' top-down BACT regulations interact
14 with the federal CAAA provisions?

15 A: The requirements under the CAAA will tend to raise the ceiling
16 on determining cost-effectiveness for emissions control
17 measures, particularly under Title I. Title I does not
18 specify the control measures that the states are to use to
19 meet air quality targets; it requires the states to submit
20 state implementation plans (SIPs) that outline how the federal
21 air quality targets and milestones will be achieved. Within

22 ³⁹CAAA Title IV, Section 407(b)(1). Low-NO_x burners are not
23 suitable for cyclone boilers, which would require something like
24 gas reburn, at a cost roughly six times higher than that of low-
25 NO_x burners. (Illinois DENR, Analysis of Acid Rain Control
26 Alternatives. October 1989).

27 ⁴⁰For instance, see United Engineers & Constructors, Edgar
28 Energy Park Project Draft Environmental Impact Report, p. 4.6-1.
29 February 16, 1990.

1 this framework, each state develops and implements its SIP,
2 presumably at lowest cost for that state given technical and
3 administrative constraints. Other titles of the CAAA (listed
4 above) set minimum NO_x and VOC control requirements, such as
5 specific NO_x emissions limits on mobile sources in Title II,
6 specific minimum NO_x control measures on stationary sources to
7 reduce acid rain effects in Title V, and minimum NO_x and VOC
8 emissions controls requirements as a member of the Northeast
9 transport region. These provisions, along with current air
10 quality regulations, set the base case for the SIP, onto which
11 more expensive controls, on new and existing sources, are
12 added until ozone air level targets defined by the SIP can be
13 achieved and permanently maintained. Massachusetts has not
14 yet revised its implementation plan to respond to the CAAA,
15 and is not required to do so until up to four years after the
16 enactment of the CAAA. However, New York estimates that
17 retrofitting of SCR on coal plants will be necessary to meet
18 CAAA Title I in that state, at a marginal cost of about
19 \$8,800/ton.⁴¹ The DPU's valuation of NO_x at \$6,250 is quite
20 reasonable in the light of New York's estimate, and may be
21 somewhat understated.

22 23 24 3.4 Particulates

25 ⁴¹New York State 1991 Draft Energy Plan Issue Report, Issue 9.

1 Q: Have you updated your estimate of the value of reducing
2 particulate emissions?

3 A: Yes. Fabric filters may be determined to be BACT control for
4 particulates in Massachusetts.⁴² Applied Energy Systems (AES)
5 determined that the incremental cost of particulate removal
6 from 0.015 lbs/MMBtu to 0.012 lbs/MMBtu emissions rate, using
7 a fabric filter on its proposed Harriman coal plant
8 (Bucksport, Maine), was \$37,260/ton particulates (1995\$) or
9 about \$31,200/ton particulates (1991\$). AES determined, based
10 on this analysis, that 0.02 lb/MMBtu was BACT for particulate
11 control using a fabric filter. However, this level of
12 emissions from a coal plant is below what is generally
13 required for new coal sources in New England. In fact, new
14 sources are generally not required to achieve 0.015 lb/MMBtus.
15 The expected emission level from the Eastern Energy New
16 Bedford Coal AFBC appear more typical for new plants, 0.02
17 lb/MMBtu. Therefore, the incremental cost calculated by AES
18 is probably higher than the costs of marginal control measures
19 required in New England at present.

20 However, AES estimate does provide some useful
21 information. If this incremental cost for very high
22 reductions is comparable to ESP costs, then the cost of
23 achieving reductions down to 0.018 lb/MMBtu would be on the
24 order of 82% that of the cost of going from 0.015 to 0.012,

25 ⁴²Personal communication with A. Aiken of NEES, 1990.

1 or \$25,584/ton⁴³. Even to achieve 0.03 lb/MMBtu emissions
2 level, the cost would be about 62%, or \$19,344/ton. These
3 costs may overstate the value of particulate reductions
4 somewhat due to differences between baghouses and ESP and the
5 company's use of nominally levelized costs. Nonetheless, it
6 seems likely that the marginal cost of controls required in
7 New England could be on the order of \$10,000/ton. The DPU
8 should consider increasing the particulate externality value.

9 An excerpt from that analysis is attached as Figure 3.4.
10 It is not clear if such a stringent emissions level would be
11 required in Massachusetts or elsewhere in New England.
12 Therefore, we continue to support the use of the DPU value for
13 particulates, which was based on our earlier analysis of the
14 marginal cost of improving ESP efficiency from 95% to 99%, as
15 a minimum value. However, in light of periodic reports in the
16 scientific community regarding previously unsuspected serious
17 health consequences of small particulate matter, we think that
18 this externality is probably undervalued.⁴⁴
19

20 ⁴³Cost = $K \times \ln(E)$, where $E = 1 - (\text{ESP efficiency})$. (See
21 workpapers for calculations)

22 ⁴⁴The value of particulate emissions is complicated by the fact
23 that some of the health effects attributed to particulates may be
24 more properly attributed to sulfates and nitrates, which are
25 emitted from the stack as SO₂ and NO_x respectively, and toxics,
26 such as metals, some of which are emitted as gases (mercury) and
27 others which adhere to particulate matter. This problem is unique
28 to direct valuation and does not affect particulate valuation based
29 on the cost of control measures.

3.5 Carbon Monoxide (CO)

Q: Do you have additional information on costs for carbon monoxide (CO) reduction?

A: Yes, we have one additional cost estimate. The Massachusetts Institute of Technology (MIT) proposed cogenerator project proposed the use of a CO catalyst to reduce its CO emissions by 90% from 148.4 tons/year at a cost of \$1,100/ton CO removed.⁴⁵ MIT asserts that the catalyst was BACT for CO control and justified for the urban location of MIT, which was in non-attainment for CO in 1987 and 1988.⁴⁶ This value might be justified for regions within Massachusetts that are out of federal attainment for CO, but may be too high for other areas. We do not recommend a change to the DPU externality value for CO of \$820/ton (\$1989) at this time.

3.6 Update on Fuel Cycle Externalities Studies

Q: What is currently being done about developing upstream fuel-cycle externalities?

A: Three federally sponsored studies look promising for providing fuel-cycle externalities for energy resources within the next few years:

⁴⁵MIT does not state whether this is a real or nominally levelized cost. However, it appears to be real levelized based on total initial capital cost of \$300,000, \$275,000 of which is for catalyst which must be replaced every two years, and \$10,000 other annual operating costs. Estimate appears to be in \$1989.

⁴⁶U.S. Department of Commerce, 1990 Statistical Abstract of the United States.

1 Oak Ridge National Lab, in conjunction with the European
2 Community and Resources for the Future, is developing life-
3 cycle emissions and effluents figures, including valuation of
4 effects, for conventional and renewable power generation
5 resources for electricity and transportation fuels where they
6 overlap. The ORNL and the EC are together currently
7 developing coal life-cycle externalities. Other fossil,
8 biomass and nuclear externalities will be developed by the
9 ORNL, while the EC has taken responsibility for developing
10 renewable life-cycle externalities. Preliminary coal results
11 are expected in a Primer this fall, after a guidance document
12 was released this summer for comment. Oil and biomass
13 externalities are expected next year. The guidance document
14 outlined an "incrementalist" perspective for the study, which
15 appears to look at incremental additions to existing
16 facilities. This approach is similar to but not the same as
17 a complete marginal perspective in that it has left out
18 several important effects, such as land use and some materials
19 required for construction. Therefore, the study's results
20 will at best provide a baseline for fuel-cycle emissions, but
21 will not completely characterize upstream externalities.⁴⁷

22 ORNL is also collaborating with Pacific Northwest Labs,
23 SERI, and the Oko Institut to develop fuel cycle externalities
24 (quantities only, no valuation) for biomass fuels and

25 ⁴⁷Personal communication with J. Beldock of the Office of
26 Environment and Energy Efficiency of the DOE.

1 renewables. This study is being coordinated by Ken Humphreys
2 of PNL. The Oko Institut (of Germany) has developed a model
3 called TEMIS (total emissions model for integrated systems)
4 for determining fuel cycle environmental effects for utility
5 resources. The model was originally developed in German using
6 data from the European community. It was translated into
7 English through funding by the Department of Energy in 1990,
8 and the original data base has been expanded to include data
9 for the United Kingdom. The model has incorporated
10 preliminary U.S. data, primarily from Argonne National Labs
11 (ANL, 1978), and will be further updated by data supplied by
12 Pacific National Labs and the Solar Energy Research Institute,
13 and possibly ORNL. Attachment 4 is a fact sheet on the Temis
14 model provided by the Oko Institut.⁴⁸

15 The DOE Office of Environmental Analysis is sponsoring
16 an input/output modelling effort for fuel-cycle externalities
17 using the INFORM model developed at the University of
18 Maryland. This study is focussing on four fuel cycles: Solar,
19 grain ethanol, a coal boiler and integrated coal gasification
20 coal plant. This study is expected to include all sector
21 effects, including, for example, cement and electronic inputs
22 to the fuel cycle.⁴⁹

23 Q: Have you updated your analysis of any upstream externalities?

24 ⁴⁸Personal communication with U. Fritsche of the Oko Institut.

25 ⁴⁹Personal communication with J. Beldock of DOE.

1 A: Yes. Regarding our estimate of the externalities of oil
2 spills, the National Research Council has published a book on
3 the methods and costs of reducing tanker spills, including
4 the cost-effectiveness of the available measures. According
5 to the NRC, in a typical year, 7,500 tons of oil is spilled
6 from groundings and collisions, the U.S. imports about 420
7 million tons of crude oil, and 600 million tons of oil is
8 moved through U.S. waters.⁵⁰ Therefore, about 0.0013% of oil
9 moved through U.S. waters is spilled. In a typical year for
10 oil spills, double-hulled tankers with hydrostatic controls
11 would cost in the range \$366,000-\$539,000 per ton of oil saved
12 (i.e., per avoided ton spilled). Therefore, the cost of a
13 double-hulled tanker with hydrostatic controls would be about
14 \$4.76-\$7.00/ton of oil moved through U.S. waters, or \$0.12-
15 \$0.17 per MMBtu of oil moved through U.S. waters.⁵¹ This
16 estimate compares closely to our original estimate of
17 \$0.20/MMBtu calculated in Chernick and Caverhill (1989).

18 Double-hulled tankers with hydrostatic controls are the
19 most expensive and most effective measure studied by the NRC.
20 Double-hulled tankers without hydrostatic controls would cost
21 about 60% of this cost. These are assumed to be the costs of
22 new vessels. The federal oil spill bill requires new vessels
23 to be built with double hulls and many existing vessels to be

24 ⁵⁰National Research Council Committee on Tank Vessel Design,
25 Tanker Spills Prevention by Design, National Academy Press, 1991,
26 pages 173-174.

27 ⁵¹Assuming 7.3 lbs/gal and 150,000 Btu/gal of crude oil.

1 retrofit with double hulls or taken out of service within 20
2 years.⁵² It was not clear from the lay press whether the bill
3 requires hydrostatic controls or not. The cost of the
4 retrofit is likely to be higher than the cost for a new
5 vessel, although we do not have an estimate of the cost.

7 ⁵²Cushman, J.H., "Oil Spill Compromise Calls for Double Hulls,"
8 New York Times. July 13, 1990.

1 4. ADDITIONS

2 4.1 Chlorofluorocarbons (CFCs)

3 Q: What regulations have been proposed or adopted concerning the
4 production and use of CFCs?

5 A: The most widely known treaty is the Montreal Protocol.
6 Last revised in June 1990, the pact calls for a 20%
7 reduction in CFC consumption by 1993 and a 50% reduction
8 by 1995. Under terms of the agreement, CFCs, halons,
9 and carbon tetrachloride (CCl₄) will be phased out
10 entirely by 2000 and methyl chloroform will be phased
11 out by 2005. Approximately seventy countries have agreed
12 to follow the Montreal Protocol's terms.

13 On the national level, several countries have adopted CFC
14 policies that are even more strict than the Montreal Protocol.
15 The CAAA legislates the elimination of CFC, CCl₄, and methyl
16 chloroform on a more accelerated basis than called for under
17 the Montreal Protocol. In addition, HCFC's are to be phased
18 out between 2015 and 2029, depending on their application, if
19 they have been recycled, and other factors. In Switzerland,
20 HCFC's are being banned between 1992 and 1994 and CFC's by
21 1995.⁵³ Germany also passed a law prohibiting CFC production
22 and use by 1995 and HCFC-22 production and use by 2000.⁵⁴

23 Q: Is there any precedent for CFC regulations at the state level?

24 ⁵³Global Environmental Change Report, Vol. III, No. 17.
25 September 6, 1991.

26 ⁵⁴Global Environmental Change Report, Vol. III, No. 17. April
27 5, 1991.

1 A: Yes. Beginning in 1995, Vermont will not allow any motor
2 vehicle using CFC's in its air conditioning to be sold in
3 state. The South Coast Air Quality Management District has
4 also adopted a CFC elimination policy which among other
5 things, commits to phase out all uses of CFCs and halons as
6 soon as possible and before 1997, and pledges to phase out all
7 HCFC uses as soon as possible.⁵⁵

8 Q: What quantity of CFC's is emitted by typical equipment?

9 A: CFC's may be emitted from equipment by servicing, leakage,
10 recycling, or simply disposal. A large commercial air
11 conditioner can emit up to 400 lbs/year of CFC's.⁵⁶ A typical
12 residential refrigerator contains approximately one pound of
13 CFC-12,⁵⁷ which would eventually be released to the atmosphere
14 if the unit is disposed of in a landfill and which would be
15 released immediately if the unit is crushed or shredded.

16 Q: How can the relative influences of different chlorinated
17 compounds be measured?

18 A: As they contribute to global warming and ozone depletion,
19 CFC's, HCFC's, CCl₄, and methyl chloroform are commonly
20 evaluated as to their global warming potential (GWP) and ozone
21 depletion potential (ODP) relative to CFC-11. Table 4.1.1,

22 ⁵⁵SCAQMD Policy on Global Warming and Stratospheric Ozone
23 Depletion, Public Workshop, February 8, 1990. This document
24 contained a draft policy. The final version was similar according
25 to personal communication with SCAQMD.

26 ⁵⁶Global Environmental Change Report, Vol. II, No. 13, July 6,
27 1990.

28 ⁵⁷ASHRAE Handbook E.37.6.

1 columns 2 and 3, show ozone depletion potentials (ODPs) and
2 global warming potentials (GWPs) for several
3 chlorofluorocarbons (CFCs), hydrochlorofluorocarbons (HCFCs),
4 and hydrofluorocarbons (HFC).

5
6 a. Costs of CFC phaseout

7 Q: Are CFCs being phased out because of their contributions to
8 both stratospheric ozone depletion and the greenhouse effect?

9 A: No. At least under the Montreal protocol, CFCs are being
10 phased out due to their contribution to stratospheric ozone
11 depletion alone. However, since these chemicals also
12 contribute to the greenhouse effect, their phaseout is often
13 credited with reductions in total greenhouse gas emissions.
14 We assume that the costs of the phaseout were justified by
15 the effects on the stratospheric ozone layer alone, and that
16 the Montreal protocol participants did not materially consider
17 the global warming benefits of CFC reductions when deciding
18 to phase out all chemicals with stratospheric ozone depletion
19 potential.⁵⁸

20 Q: What are the costs of the phaseout?

21 A: The National Academy of Sciences (1991) reports abatement
22 costs for the elimination of CFC uses in the U.S. According
23 to the NAS, the most expensive measures required under the
24 phaseout are replacement of refrigerants and insulation in

25 ⁵⁸Essentially, this makes the global warming benefits of CFC
26 reduction free goods, and the costs of CFC control imply nothing
27 about the value of reducing CFCs for global warming purposes.

1 appliances with refrigerants and insulations that contain
2 hydrofluorocarbons (HFCs).⁵⁹ The NAS (Table J.1) reports that
3 the cost of fluorocarbon substitutes for appliance insulation
4 is \$72,429/tonne of CFC replaced (at a carrying charge of
5 10%).⁶⁰ Since the HFC replacement has an ODP of zero, the
6 ozone depletion potential of the appliance insulation is
7 reduced to zero. Therefore, if CFC-11 is the replaced
8 insulator, then the cost of eliminating the ozone depletion
9 potential of appliance insulation is \$72,429/tonne CFC-11, or
10 \$65,700/ton CFC-11.

11 Similarly, if CFC-11 is replaced with HFC in appliance
12 refrigerants, the NAS reports a cost of \$74,500/tonne CFC-11
13 (at a carrying charge of 10%) or \$67,600/ton CFC-11.⁶¹

14 Q: What do these costs imply about the value of reducing CFC
15 emissions?

16 A: To estimate the value for the reduction of CFC emissions for
17 utility planning, especially the value of capturing CFCs in

18 ⁵⁹National Academy of Sciences, "Policy Implications of
19 Greenhouse Warming Report of the Mitigation Panel," Tables J.1 and
20 J.2, Cost Impacts of CFC phaseout - United States and Worldwide are
21 attached as Figures 4.1.1 and 4.1.2. For the calculation presented
22 here the U.S. costs are used; Worldwide costs are higher for the
23 same measures.

24 ⁶⁰A tonne is one metric ton. In the NAS report they use "ton"
25 to mean metric ton.

26 ⁶¹CFC-12 is also a common refrigerant in this application. If
27 the NAS assumed CFC-12 was replaced in its cost calculation, the
28 value for CFC-11 equivalent would be slightly higher due to the
29 slightly lower ODP of CFC-12. For both applications, if the
30 replacement chemical had not had an ODP of zero, the net reduction
31 in ODP would have been lower for the same cost, and the cost of the
32 measure would have been higher per ton of CFC.

1 DSM program design, the cost of replacing appliance
2 refrigerants provides the best estimate. Although the
3 Montreal protocol and the CAAA require essentially a ban on
4 all new CFC uses, they do not require more expensive CFC
5 emissions reducing measures, such as capture and destruction
6 of CFCs in many existing uses. Clearly, the measures included
7 in the CFC regulation have been individually considered for
8 feasibility and cost-effectiveness.⁶² Therefore, we estimate
9 that the value of reducing CFC emissions is implied by the
10 cost of replacing CFCs with HFCs in appliance refrigerants,
11 and is \$67,600/ton CFC-11 (\$1990).

12 Q: What are the values of reducing the other ozone-depleting
13 chemicals?

14 A: To calculate the value of reducing emissions of other CFCs and
15 HCFCs, we can use the relative ODPs of the chemicals. Table
16 4.1.1, column 4, shows externality values of the ozone
17 depleting chemicals.

18 Q: How should these values be used in resource selection?

19 A: Resources that reduce CFC emissions should receive a credit
20 of \$67,600/ton CFC-11 equivalent reduced. DSM programs that
21 collect and recycle or destroy CFCs should get a credit for
22 the CFC emission reduction. For example, a DSM program that
23 offers refrigerator retirements including proper CFC
24 destruction should receive this credit. A refrigerator

25 ⁶²Indeed, when the Montreal Protocol was adopted, the cost
26 estimates for many measures were much higher.

1 retirement program that retires refrigerator coolant CFCs
2 without release to the atmosphere is worth \$34 per
3 refrigerator for the CFC recovery alone.⁶³

4 5 4.2 Air Toxics

6 Q: What are air toxics, and why do they matter?

7 A: Combustion of oil and coal release a variety of toxic
8 substances into the air, which include organics, such as
9 formaldehyde, and heavy metals such as arsenic and beryllium.

10 The amount emitted from power plants varies considerably
11 depending on many factors, including the type of fuel, the
12 extent of preprocessing, which is usually intended to remove
13 other materials, such as sulfur, the type of equipment used
14 for SO₂ and particulate control (e.g., precipitators,
15 baghouses, scrubbers), and operating conditions. Metals occur
16 naturally in coal and crude-oil deposits. Following
17 combustion, some portion of each metal is released with the
18 flue gases, while the remainder ends up in fly ash, bottom
19 ash, or air-pollution control equipment. Toxic trace metals
20 tend to be found in highest concentrations in the smallest
21 particles emitted from the stack. Such small particles are
22 especially harmful because they easily penetrate deep into the
23 lungs, where they can enter the bloodstream, and they are
24 deposited in bodies of water where they bioaccumulate in fish.

25 ⁶³Calculation is \$67,600/ton x 1/2000 x 1 lb/refrigerator =
26 \$34/refrigerator.

1 Q: How have you approached the problem of estimating the value
2 of reduced emissions of air toxics?

3 A: We recognized that it would not be possible to develop
4 marginal costs of control for each individual pollutant.
5 Table 4.2.1, reproduced from an EPRI paper,⁶⁴ lists 36 toxic
6 air pollutants that have been found in power plant flue gas.
7 As shown in Tables 4.2.6 and 4.2.7, we have found data on
8 power plant emissions of over a dozen of these air toxics,
9 mostly heavy metals.

10 All of the listed materials, other than copper, are
11 listed as hazardous air pollutants in Title III of the CAAA.
12 The EPA is to set maximum achievable control technology (MACT)
13 for each of these 14 categories, and 175 others, over the
14 course of the decade. Each source that emits over 10
15 tons/year of any one toxic, or over 25 tons/year of all toxics
16 combined, will be covered.⁶⁵ A power plant would be a single
17 source under the definition of "source." In addition, a
18 special study of utility power plant toxics emissions must be
19 completed by 1993 and used as the basis for regulation of

20 ⁶⁴Chow, W. et al., Managing Air Toxics. 1990.

21 ⁶⁵Given the large amount of chlorine emitted by coal plants,
22 and the large amount of nickel emitted by residual oil plants,
23 virtually all utility power plants using these fuels would be
24 covered by the size limits. For example, even at just a 30%
25 capacity factor, a residual plant of more than 600 MW would exceed
26 the 10-ton limit for nickel. The limits would be exceeded by
27 smaller plants burning coal, or operating at higher capacity
28 factors. Since a source is defined as including a "group of
29 stationary sources located within a contiguous area and under
30 common control," the limit would apply for power plants, not for
31 units.

1 power plant emissions.⁶⁶ Thus, control requirements for these
2 emissions will be changing rapidly over the next several
3 years, and regulation of utility emissions may lag behind that
4 of emissions from other sources.

5 Hence, we have concentrated on relating the toxicity of
6 the various toxics to one another, and on estimating the costs
7 of required controls for benchmark toxics.

8 Q: How have you determined the relative toxicity of the various
9 air toxics?

10 A: There is no fundamental unit of toxicity (such as mortality
11 per gram) that is applicable to all toxics. Different toxics
12 have different effects, including carcinogenicity,
13 neurological effects, retardation of mental and physical
14 development, and damage to a number of body systems (e.g.,
15 kidneys, lungs, blood). Each individual toxic may be known
16 or suspected to have several effects, and each of those
17 effects may be subject to considerable uncertainties. Hence,
18 to relate the toxicity of the various materials, we used the
19 Threshold Effects Exposure Levels (TELs),⁶⁷ the maximum air

20 ⁶⁶The Clean Air Act Amendments appear to defer application of
21 the air toxics provisions to electric utilities, even though
22 arsenic emitted by a power plant is as dangerous as arsenic emitted
23 by a smelter. The apparent rationale for the exemption is that the
24 utilities will be responsible for most of the acid rain reduction,
25 and in some cases for much of the local air quality improvements,
26 mandated in other sections of the same bill, and that equity
27 requires that utilities be temporarily exempted from the burdens
28 of complying with the air toxics provisions.

29 ⁶⁷The TELs are the Massachusetts version of 24-hour limits of
30 ambient concentrations, more generally known as allowed air levels
31 (AALs).

1 concentrations set for the general environment by the
2 Massachusetts Department of Environmental Protection. These
3 are shown in column 2 of Table 4.2.2. The lower the TEL, the
4 higher the DEP's assessment of the material's danger to the
5 public, per gram in the air, considering the health effects
6 and the uncertainties in those effects. Column 5 of Table
7 4.2.2 shows the ratio of the TEL for lead (Pb) to that of the
8 particular toxic. This ratio is a measure of the toxicity of
9 the material, with respect to lead.

10 For example, the lowest TEL is that of beryllium, which
11 is 140 times lower than the TEL for lead. Hence, a microgram
12 of beryllium pollutes 140 times as much air to its TEL as does
13 a microgram of lead. At the other extreme, 14.5 times as much
14 HCl as lead is required to reach the TEL, so hydrogen chloride
15 is one fourteenth as toxic as lead.

16 For the pollutants for which the DEP has not set TELs or
17 AALs (arsenic, manganese, and polycyclic organic matter, or
18 POM), we determined column 5 from the ratio of AALs set by
19 other states. For arsenic and manganese, we used the AALs of
20 Connecticut, the only other New England state for which the

1 necessary AAL have been reported.⁶⁸ For POM, we used Virginia,
2 the only state that has set an AAL.

3 Q: How did you estimate the marginal costs of control of air
4 toxics?

5 A: We looked at a number of figures, based on required or
6 anticipated controls.

7 Arsenic from primary copper smelters: Table 4.2.3 shows
8 the costs of the ESP required at that ASARCO-El Paso copper
9 smelter to reduce arsenic emissions, along with the costs of
10 controls at other facilities, which are not currently required
11 for these existing facilities. The cost of the control is
12 \$18/lb, in roughly 1982\$. Given the high relative value of
13 arsenic compared to lead, this is less than \$1/lb Pb, and is
14 thus not marginal.

15 EEA (1990) indicates that the CAAA will require arsenic
16 controls on all remaining primary copper smelters. From EPA
17 (1986) and Table 4.2.3, we see that these controls cost as
18 much as \$855/lb in the dollars of the study in 1982\$.⁶⁹ Thus,

19 ⁶⁸As we will see later, the relative environmental cost of As
20 and Pb is important, since As is a major contributor to total power
21 plant toxic emissions. The AAL's for As set by Nevada and Virginia
22 (the other two states that set AALs for both Pb and As) are roughly
23 two orders of magnitude higher than those set by Connecticut, even
24 though all three states use similar AALs for Pb. If Massachusetts
25 adopted the relative valuations for As used by Nevada and Virginia,
26 the total lead-equivalent emissions from oil and coal plants would
27 be considerably lower. On the other hand, the observed costs of
28 control for As would become significant in terms of \$/lb Pb.

29 ⁶⁹The dated cost data in the study are stated in 1982\$. Due
30 to the uncertainties in relative toxicity, a few percent of
31 inflation is not a significant uncertainty.

1 these controls would cost about \$1200/lb As in 1991\$. Since
2 EEA (1990) projects much larger As reductions than does EPA
3 (1986) (about 300 tons/yr, as compared to 35 tons/yr), EEA
4 (1990) may also assume other, more expensive, controls. The
5 incremental level of control for each application (e.g., the
6 decision to capture 85% of emissions rather than 80%) will
7 often be still more expensive.

8 Arsenic from glass manufacturing plants: EPA (1986)
9 estimates the costs and effectiveness of adding electrostatic
10 precipitators (ESPs) to glass furnaces, to control As
11 emissions. Of the furnaces for which controls were required
12 at that time, the most expensive control was for a furnace
13 that emitted 3.75 tons/yr. The control was expected to cost
14 \$532,400/yr (roughly 1980\$) and the reduction was required to
15 be at least 85% of the emissions, or 3.19 tons/yr. The cost
16 of control was thus \$84/lb As in 1980\$, or about \$125/lb As
17 in 1991\$.⁷⁰

18 The requirement to achieve 85% reduction, as opposed to
19 some other lower level of reduction, has a higher incremental
20 cost. From Radian (1984, p. 52), the size (S) of the ESP
21 required to remove a fraction (F) of particulates follows the
22 formula:

$$S = -k * \ln(1 - F)$$

24 where k is a constant. For F = .85, S = 1.9 k; for F = .80,
25 S = 1.6 k. Thus, roughly 16% (.3/1.9) of the system cost is

26 ⁷⁰New furnaces will be subject to stricter limits.

1 due to the last 6% (5%/85%) of collection efficiency. Thus,
2 the incremental cost is about 2.7 (16%/6%) times as high as
3 the average cost. This brings the incremental cost of the
4 required reduction to about \$300/lb As. So long as arsenic
5 is valued at 60 times the environmental cost of Pb, this is
6 still only about \$5/lb Pb.

7 Lead from secondary lead smelters: EEA (1990), in a
8 study for EPA, estimated the cost of controls expected to be
9 required of secondary lead smelters under the Clean Air Act
10 Amendment. These controls include the installation on process
11 stack sources of venturi scrubbers downstream of the baghouse,
12 enclosure of the smelter building with ventilation to a new
13 baghouse, use of a water sprinkler system to suppress fugitive
14 dust from area emission sources, and replacement of the blast
15 furnace with a new electrolytic process. The analysis
16 estimates the average cost of these controls to be
17 \$330,415/ton Pb, or \$165/lb Pb(1989\$). Given the variation
18 in costs between plants, the most expensive controls are
19 likely to be several times as expensive, or roughly \$500/lb
20 Pb.

21 Chromium in cooling towers: EPA (1988, Table 7-1) found
22 that replacement of chromium water-treatment products with
23 phosphate water treatment would be cost-effective, even though
24 it could cost up to \$1,170,000/ton Cr nationwide, and as much
25 as \$4,270,000/ton Cr for the 2,780 smallest cooling towers
26 considered. The middle of the cost range for the small

1 cooling towers is \$2,210,000/ton Cr. All of these costs
2 appear to be in 1986\$. The middle of the small-tower cost
3 range is thus about \$1330/ton Cr, in 1991\$.

4 The particular type of chromium used in cooling tower
5 corrosion control is hexavalent chromium (Cr^{+6}), which is
6 particularly carcinogenic. The only state to report an AAL
7 for hexavalent chromium, Connecticut, sets that AAL an order
8 of magnitude lower than the AAL for chromium in general.
9 Massachusetts rates chromium (TEL of 1.36) as almost an order
10 of magnitude less toxic than lead (TEL of .14). Thus, the
11 cost of the hexavalent chromium controls are roughly
12 equivalent to about \$1300/lb Pb.

13 Lead in paint: One of the most clearly documented and
14 expensive efforts to control heavy metals is the effort to
15 remove lead from paint.⁷¹ Regulations often require removal
16 or encapsulation of paint containing more than 1 gram Pb per
17 cm^2 in buildings that are or might be occupied by children.⁷²
18 The costs of this control can vary widely, depending on the
19 care taken to minimize lead dust, the cost of labor, and other
20 considerations. Since lead control may overlap with repair
21 of degraded surfaces, repainting, repapering, and other

22 ⁷¹For example, see U.S. Department of Housing and Urban
23 Development, Comprehensive and Workable Plan for the Abatement of
24 Lead-Based Paint in Privately Owned Housing.
25
26

27 ⁷²Pollack, S., "Solving the Lead Dilemma," Technology Review.
28 October 1989.

1 structural and aesthetic improvements, identifying specific
2 lead-control costs can be complicated.

3 Fortunately, we do not need to characterize the entire
4 range of lead paint removal techniques, but only the marginal
5 costs. One of the more expensive control measures involves
6 deleading of windows. The small, uneven surfaces of mullions
7 (the dividing bars between the lights of windows) and other
8 window trim are difficult to scrape, so the least-cost option
9 for deleading windows is generally to replace the entire
10 window. Replacing just the sash (the movable portions of the
11 window) costs about \$15/ft², while it costs about \$25/ft² to
12 replace the entire window (including the frame).⁷³ Both the
13 surface area of the frame and the costs of alternative
14 treatments (such as scraping) will vary. To simplify the
15 analysis, we concentrated on the cost of lead abatement
16 through replacement of window sash.

17 The cost of sash replacement per gram of lead will vary
18 with the number of lights (which increase the area of the
19 mullions and hence the amount of lead), the size and shape of
20 the window, and the concentration of lead. We examined a
21 number of windows, and found that 3600 cm² of painted area was
22 fairly typical. One could get about that much painted area
23 from such representative cases as:

24 ⁷³Personal communication from Blair Hamilton, Vermont Energy
25 Investment Corporation, based on experience of conservation
26 programs in New England.

- 1 • a 4-over-4-light window of 2x4', with about 2" of exposed
- 2 cross-section on each mullion and rail (for 24'*2" = 4
- 3 sq. ft. of area), or
- 4 • a 6-over-1 window of 30" by 62", described in greater
- 5 detail in the workpapers.

6 Thus, at 1 mg/cm², it would not be unusual to replace 8 to 13
7 sq. ft. of window to eliminate 3.6 g of lead. At \$15/ft²,
8 this would cost \$120 - \$200, or \$33 - \$54/g Pb. Since there
9 are 454 gram in a pound, the cost of window replacement for
10 lead abatement is \$15,000 - \$25,000/lb Pb. Public agencies
11 require, and in some cases pay for, lead abatement at this
12 level of cost-effectiveness.

13 Air-borne lead is obviously different than lead in paint.
14 Air-borne lead is more important than lead in paint, since all
15 air-borne lead is already in the environment, while most of
16 the lead in paint will stay bound indefinitely. On the other
17 hand, lead in paint, particularly interior paint, can create
18 "hotspots" of lead contamination that can have devastating
19 effects on the physical and mental development of individual
20 children, while the airborne lead will be more evenly
21 distributed across the population of humans and other species.
22 There does not seem to be any straightforward way to convert
23 the value of lead reduction in paint to the value of reducing
24 air-borne lead emissions. Nonetheless, the high implied value
25 of removing lead in paint, combined with the small amounts

1 known to affect human development, suggests a very high value
2 per pound for air-borne emissions as well.

3 Relative exposure limits for air toxics, PM, and SO₂:
4 Table 4.2.4 extends the approach of Table 4.2.2 to relate air
5 toxics to other regulated air pollutants, which (such as air
6 toxics) are regulated due to their direct health effects.⁷⁴
7 Much higher exposure levels are permitted for PM and SO₂ than
8 for the air toxics, implying that the equivalent health
9 effects (reflecting the differences in the types of health
10 effects, and the uncertainties) of the toxics are
11 correspondingly higher than those of PM and SO₂. Given the
12 relative health valuation, and the DPU's current valuation of
13 SO₂ and PM, each pound of lead is worth about \$2000.

14 Q: Please summarize your estimates of the value of lead
15 emissions.

16 A: Table 4.2.5 lists the valuations from the analyses described
17 above. Recall that the control costs represent minimum,
18 rather than maximum, values.

19 Q: How have you used this range of estimates?

20 A: We can be quite confident that airborne lead has a social cost
21 well in excess of \$150/lb, from the cost of smelter controls.
22 The cost of chromium controls in chiller cooling towers
23 suggests that the value of reducing lead emissions is at least
24 on the order of \$1,500/lb. A somewhat higher value is

25 ⁷⁴NO_x and VOCs are regulated primarily due to their
26 contribution to forming ozone, and are thus omitted from this
27 analysis.

1 supported by the relative stringency on exposure limits to
2 lead, PM, and SO₂, given the DPU's current valuation of those
3 pollutants. The cost of replacing lead-painted windows
4 suggests a social cost of lead in excess of \$15,000/lb.

5 In Table 4.2.6, we show the effect of applying these
6 three costs to the toxic emissions of coal- and oil-fired
7 power plants.⁷⁵ Table 4.2.7 extrapolates the effectiveness of
8 ESPs, as estimated by Radian (1989), from coal plants to
9 residual-oil plants. Even at the geometric mean of the range
10 (\$1,500/ton Pb), the toxic emissions of

- 11 • unscrubbed coal plants with electrostatic precipitators
12 (ESP) would be worth about 12¢/kWh,
- 13 • scrubbed coal plants (none of which now exist in New
14 England) would be worth about 2¢/kWh,
- 15 • uncontrolled residual oil plants would be worth about
16 5¢/kWh,
- 17 • distillate-fired plants (mostly combustion turbines)
18 would be about 2¢/kWh, and
- 19 • a residual plant with an ESP would be worth about 1¢/kWh.

20 These values could easily be ten times as large.

21 These results are sensitive to a number of factors.
22 First, there are uncertainties in the relative valuation of
23 the various toxics. This is particularly troublesome in the

24 ⁷⁵This table includes the air toxics for which we had emissions
25 data and relative valuations. A number of other air toxics are
26 emitted by coal and/or oil, including fluorine, many metals, and
27 some other organics. As additional data becomes available, we will
28 supplement this table.

1 case of arsenic, which is valued from Connecticut, rather than
2 Massachusetts, regulations, and which is the most important
3 contributor to the valuations for coal and residual.⁷⁶ If As
4 were only 6 times as toxic as Pb, rather than the 60 we
5 derived from Connecticut's regulations, the total valuation
6 would fall about 25% for ESP coal, about 60% for scrubbed
7 coal, and about 30% for residual.

8 Second, emissions will vary between plants in each
9 category. For example, some ESPs and baghouses may be more
10 efficient than the ESPs EPA studied.

11 Third, the high values of reductions in air toxics and
12 the requirements for toxics controls under the CAAA may result
13 in retrofit of scrubbers on coal plants and ESPs on residual
14 plants.

15 Q: How do you recommend that the DPU incorporate air toxics in
16 the valuation of externalities?

17 A: At this point, the most important determination is for the
18 uncontrolled residual-fired plants that make up most of the
19 NEPOOL marginal supply, since near-term resource decisions
20 seem to be likely to be backing out existing sources for some
21 time. We suggest a very modest initial valuation of 1¢/kWh
22 for the air toxics from these units. We would suggest an air
23 toxics value for about 2.5¢/kWh for ESP coal plants (for
24 purchases from New York or Ontario, for example), about 4

25 ⁷⁶Chlorine is almost as important as As for ESP coal plants.

1 mills/kWh for scrubbed coal and distillate, and 2 mills for
2 ESP-equipped residual plants.

3 4 4.3 Thermal Pollution

5 Q: What do you mean by thermal pollution?

6 A: We refer here to the transfer (rejection) of heat from a power
7 plant to a body of water. All steam plants and combined cycle
8 plants must reject heat from their condensers, to maintain a
9 temperature and pressure differential across the steam
10 turbine. The cooler the cooling water, the more efficient the
11 steam turbine can be. Traditionally, New England power plants
12 have rejected heat to natural bodies of water, such as rivers
13 and harbors.

14 Q: How have you estimated the cost of this heat rejection to
15 bodies of water?

16 A: It is our understanding that the use of natural bodies of
17 water as the heat sink for a power plant (called once-through
18 cooling) is no longer permitted for most applications in New
19 England.⁷⁷ Other than proposals for repowering or reusing
20 existing utility power plant sites (at Edgar and Manchester
21 Street), we are not aware of any serious recent proposal to
22 use once-through cooling in New England. In other words,
23 environmental regulators have essentially determined that the
24 environmental costs of once-through cooling (e.g., pulling

25 ⁷⁷Once-through cooling became rare in some other parts of the
26 country in the 1970s.

1 small organisms through the cooling systems, injuring fish and
2 other large organisms through impingement against the intake
3 structure) exceed the costs of requiring other cooling
4 methods.

5 Most new power plants use wet cooling towers for heat
6 rejection. Wet cooling towers use evaporation of water to
7 remove the heat. In the process, they consume fresh water,
8 usually release concentrated pollutants (and water-treatment
9 chemicals) in the "blowdown" of unevaporated water,⁷⁸ produce
10 plumes of water vapor (which can result in local fogging,
11 visibility problems, and icing of roads in the winter), and
12 release to the air whatever toxics are present in the water.
13 A few plants use more expensive and less efficient dry cooling
14 systems, which are similar to very large automobile radiators.
15 Dry cooling systems are environmentally beneficial, since they
16 release nothing but heat to the air, but they are more
17 expensive to build and operate, and increase the plant's heat
18 rate even more than do wet cooling towers. Dry cooling towers
19 are not generally required.

20 We can estimate the environmental costs of once-through
21 cooling from the extra costs regulators are willing to require
22 to avoid it. For this purpose, we will use the cost
23 differential between once-through and evaporative cooling.
24 Since the environmental cost differential must be larger than

25 ⁷⁸Blowdown is required to limit the buildup of impurities in
26 the cooling water.

1 the direct cost differential (or else evaporative cooling
2 would not be required), and since evaporative cooling has
3 substantial environmental costs, the total costs of once-
4 through cooling must be larger than the difference in cost
5 between once-through cooling and wet towers.

6 Q: For what plants do you have estimates of the incremental costs
7 of wet cooling?

8 A: As we mention above, most power plant proposals do not even
9 discuss the possibility of once-through cooling. The
10 exceptions are Manchester Street, which has been conditionally
11 licensed with once-through cooling, due to space constraints
12 and the lack of adequate fresh water supply for evaporative
13 cooling; and Edgar, which is seeking a similar license. Both
14 New England Electric System and Boston Edison have presented
15 estimates of the additional cost of using evaporative cooling.
16 For Edgar, we have estimates of the costs of evaporative
17 cooling both at the Edgar site and at the alternative
18 Ironstone site. Table 4.3.1 presents those estimates and
19 computes the cost of control per avoided MMBTU rejected to
20 water.⁷⁹ Depending on the estimate, the real-levelized cost
21 of control is about 30-80¢/MMBTU rejected. Costs at other
22 sites may be higher. Hence, it is reasonable to assume that

23 ⁷⁹The estimates are not well documented, and Table 4.3.1
24 represents our best attempt to apply the available data. Perhaps
25 Massachusetts Electric and Boston Edison can improve on our
26 interpretation of the data.

1 the general prohibition on once-through cooling is based on
2 an implicit value of heat rejected of more than 80¢/MMBTU.

3 Q: How does this value translate to ¢/kWh?

4 A: For a gas-fired combined-cycle plant with once-through cooling
5 (Edgar or Manchester St.), rejected heat is about 2.2
6 MMBTU/MWH. At 80¢/MMBTU, this would add 0.2¢/kWh to the
7 environmental costs of the plant.

8 For a typical steam plant, using 10,000 BTU/kWh or 10
9 MMBTU of fuel per MWH, of which 3413 BTU/kWh (or 3.4
10 MMBTU/MWH) goes out as electricity (by definition) and about
11 20% goes up the stack, rejection to water is about 4600
12 BTU/kWh or 4.6 MMBTU/MWH. At 80¢/kWh, this would add about
13 0.4¢/kWh.

14 4.4 Oil Import Premium

15 Q: What is an oil import premium?

16 A: An oil import premium is a value applied to domestic oil
17 consumption to reflect the external costs of oil imported into
18 the United States caused by the national vulnerability to
19 problems of energy security and high oil prices. The oil
20 import premium reflects the costs of imported oil not
21 reflected in the price of oil, or the external costs. In the
22 terminology used by the DPU in its decision in DPU 89-239, the
23

oil import premium would be considered an economic externality rather than an environmental externality.⁸⁰

Q: How does it relate to the subject of this docket?

A: In its Vote to Open Investigation (DPU 91-131, June 14, 1991), the DPU addresses the periodic update of externality values. The language of the Vote to Open Investigation repeatedly uses the phrase environmental externalities; however, the scope was not expressly limited to any particular externalities, except as outlined in the guidelines referred to in DPU 91-131 (at 4) and contained in DPU 91-141 (pages 23-24).

Department consideration of an oil import premium is consistent with these guidelines. The first guideline directs proposals to avoid externalities that are accounted for elsewhere in the resource selection criteria or the siting process. The oil import premium is not considered elsewhere in the resource selection criteria or the siting process.⁸¹

The second guideline directs proposals to focus on

⁸⁰In DPU 89-239 (at 81, footnote 37) the Department raised the question of whether the oil import premium should be included in resource selection as a monetized externality or as an issue related to fuel diversity. The DPU deferred decision on this matter, but encouraged the electric companies to address this issue explicitly.

⁸¹The oil import premium is not a utility fuel diversity issue. An oil import premium reflects the notion that all reductions in domestic use of oil reduce the external costs borne nationally by our increase in imported oil dependence. Fuel diversity issues typically reflect the specific attributes of a particular utility's resource mix. For some utilities it may be advantageous from a fuel diversity perspective to increase their fraction of oil resources in their resource mixes, even though this would increase national dependence on imported oil.

externalities that have global or regional impacts. The oil import premium reflects an externality affecting the nation as a whole. The third guideline outlines the acceptable estimation techniques. The oil import premium estimated in Chernick and Caverhill (1989) was based on the DPU's preferred method of estimation, direct valuation of effects. Therefore, the Department should reconsider the adoption of an oil import premium.

Q: What was the value of the oil import premium estimated in Chernick and Caverhill (1989) and discussed in DPU 89-239 (at 81)?

A: Chernick and Caverhill (1989) estimated an oil import premium of \$2.26/MMBtu of fuel input, in 1988\$. The discussion of the derivation of this value from Chernick and Caverhill (1989) is attached as Figure 4.4.

Q: Are you resubmitting this estimate for Department consideration?

A: Yes. This estimate is still relevant for the value of reducing the external costs of domestic oil use related to national security and price risk linked to oil imports. The Department should consider including this externality in the utilities resource-selection process.

1 5. OTHER STATE REGULATORY ACTIONS

2 Q: Are other states active in valuing externalities?

3 A: Several states have initiated proceedings or have come out
4 with Orders adopting explicit valuation of externalities since
5 the DPU's IRM Order in DPU 89-239, which included explicit
6 valuation of externalities. Prior to the DPU Order in 89-
7 239, New York was the only state that had adopted a monetized
8 externality policy. Since August of 1990, when the DPU
9 adopted Order DPU 89-239, the California Energy Commission,
10 the California Public Utility Commission and the Nevada Public
11 Service Commission have adopted specific externality values;
12 the Bonneville Power Association has proposed externality
13 values; and New Jersey has adopted externality adders for
14 electric and gas conservation program screening. These recent
15 orders are summarized in Table 5.1. Generally, these states
16 are also engaged in integrated resource planning proceedings,
17 and are considering externalities as a part of the IRP (or a
18 similar) process.

19
20 5.1 California

21 Q: Have there been any further developments in the valuation of
22 externalities before the California Energy Commission and
23 California Public Utilities Commission, since your testimony
24 in DPU 89-239?

1 A: The California Energy Commission (CEC) adopted externality
2 values for six air pollutants in 1990.⁸² In general, the CEC
3 relied on two sources of valuation estimates for air
4 emissions: the CEC staff and JBS Energy (whose work was on
5 behalf on the Independent Energy Producers (IEP)). The CEC
6 adopted values developed by JBS Energy for the out-of-state
7 NO_x and SO_x values, and the staff for the other in- and out-
8 of-state values. The staff approximated the marginal cost of
9 control in the South Coast Air Quality Management District
10 (SCAQMD), which surrounds Los Angeles and is the most polluted
11 area of California, for each pollutant by estimating the
12 average cost of a subset of the very stringent control
13 measures required in this area. The derivation of these
14 figures is in Therkelsen (1989). The result was a value lower
15 than the marginal cost of control in SCAQMD, but perhaps
16 typical of the state as a whole. The out-of-state values the
17 CEC adopted for PM-10 and VOCs were proposed by the staff, and
18 are simply 1/10th the value proposed for in-state. The lower
19 value was justified by the general federal ozone level
20 attainment status of the out-of-state regions that supply
21 power to California. The SO₂ out-of-state value developed by
22 JBS Energy and adopted by the CEC was based on mid-range
23 scrubber costs on large power plants, while the NO_x value was
24 based on JBS's estimate of selective catalytic reduction (SCR)

25 ⁸²California Energy Commission, 1990 Electricity Report.
26 October 1990.

1 costs for NO_x control on gas-fired cogenerators of unspecified
2 size. These estimates are briefly discussed in Schilberg et
3 al. (1989).⁸³ The CO₂ value used within and out-of-state was
4 developed by the staff based largely on the costs and energy
5 savings from planting trees for shading benefits estimated in
6 Akbari (1988).⁸⁴ All of the utilities are required to use the
7 same externality values.

8 The Public Utilities Commission (PUC) recently adopted
9 the externality values adopted by the CEC for emissions within
10 California,⁸⁵ but with some changes. The PUC initially
11 requires the utilities to use the CEC in-state externality
12 values, for both in-state and out-of-state emissions. The PUC
13 also ordered the utilities other than Southern California
14 Edison (which serves primarily the SCAQMD) to develop marginal
15 control cost estimates from their respective dominant air
16 basins, and to use those values for all emissions.⁸⁶

18 ⁸³Schilberg, G.M. et al., Valuing Reductions in Air Emissions
19 and Incorporation into Electric Resource Planning. August 1989.

20 ⁸⁴Akbari, H., et al., "The Impact of Summer Heat Islands on
21 Cooling Energy Consumption and CO₂ Emissions," 1988 ACEEE Summer
22 Study on Energy Efficiency in Buildings. ACEEE; Berkeley, Calif.:
23 1988.

26 ⁸⁵California Public Utilities Commission, Phase 1B Opinion:
27 Changes to Final Standard Offer Four for use in Conjunction with
28 the 1990 Electricity Report, pages 29-33. June 5, 1991.

29 ⁸⁶The PUC estimates that the values will be similar, and
30 possibly higher, than those developed for the SCAQMD and adopted
31 by the CEC.

5.2 Nevada

Q: How has the Nevada Commission addressed the issue of externalities?

A: In February the Nevada Public Service Commission adopted externality values based on a study by Tellus, which uses the cost-of-control approach. The unit values are equivalent to those of Massachusetts, inflated to 1990\$, with the following exceptions. The value for VOCs is lower to reflect costs of controls for reducing fugitive VOC emissions from gasoline in Nevada, which is generally in attainment for ozone. The values for the greenhouse gases CO₂ and CH₄ do not appear to be inflated to 1990\$.⁸⁷ Nevada is also attempting to value H₂S (which is a major emission of some geothermal plants) and requires the utilities to determine values for the site-specific externalities water impacts and land use.

5.3 New Jersey

Q: How has the New Jersey Commission addressed the issue of externalities?

A: On September 25, 1991 the New Jersey Board of Regulatory Commissioners (formerly the Board of Public Utilities) adopted adders to reflect the environmental benefits of conservation.⁸⁸ These values are apparently based on the estimates in

⁸⁷This appears to be a rounding error, since the value for the greenhouse gas N₂O was inflated to 1990\$.

⁸⁸State of New Jersey Board of Regulatory Commissioners, Conservation Incentive Rule, N.J.A.C 14:12. September 25, 1991.

1 Ottinger, et al. (1990).⁸⁹ For electric utility DSM programs,
2 the average environmental externality value is 2.0 cents/kWh.⁹⁰
3 For gas utility DSM programs, the environmental externality
4 value is \$0.95/MMBtu. These externalities are expressed in
5 1991 dollars and are to be adjusted annually with the GNP
6 deflator index. The utilities are to use externalities in
7 setting avoided costs and valuing power purchases, both in
8 bidding and negotiation.

9
10 5.4 New York

11 Q: How has the New York Public Service Commission addressed the
12 issue of externalities?

13 A: The New York Public Service Commission (PSC) relied partially
14 on work done by the New York State Energy Office (SEO) in its
15 1989 state energy plan to develop externality values for use
16 in utility planning. In its 1989 energy plan, the SEO
17 estimated the costs of various pollution control measures
18 available in that state. As discussed in Putta (1990), the
19 control costs used by the PSC to develop externality values
20 reflected a mix of high and low cost measures, except for CO₂,
21 which was arbitrarily set at a small fraction (20%) of the

22 ⁸⁹Ottinger R., et al., Environmental Costs of Electricity, Pace
23 University Center for Environmental Legal Studies, Oceana Press,
24 1990. Study is referred to as "Pace University values" in Table
25 5.1.

26 ⁹⁰This is an average value. The Commission asserts that the
27 externalities should be time differentiated by rating period but
28 does not indicate how this should be done.

1 SEO's original estimate of low-cost tree-planting. Land and
2 water use externalities were loosely based on reports prepared
3 in the Northwest for the Bonneville Power Administration.

4 The NYSEO has recently released a new draft state energy
5 plan (1991 biennial update), which includes updated control
6 costs for SO₂ and NO_x, and the costs of various CO₂ reduction
7 measures and costs of achieving emissions reductions targets.⁹¹
8 This report shows higher control cost requirements for SO₂, NO_x
9 and CO₂ for New York emissions reduction targets than
10 previously reported.⁹²

11
12 SO_x: The NYSEO estimates the external benefit of SO₂
13 reductions at \$2200/ton at the level of emissions covered
14 by the state's allowances for the year 2000.⁹³ This
15 emission level is 26% below 1988 emissions. NYSEO
16 assumes that the value of SO₂ emissions would fall
17 linearly to zero for emissions below 36% of the Clean Air
18 Act Amendments of 1990 (CAAA) allowance level. Plotting
19 the increasing cost of the supply of reductions against
20 the assumed declining benefit, NYSEO estimates the

21 ⁹¹A summary of their estimates is provided in the 1991 Draft
22 State Energy Plan biennial update, Issue 9, Table 11.

23 ⁹²The costs in the State Energy Plan were apparently calculated
24 from nominal ratemaking costs provided by the utilities, real-
25 leveled using a social discount rate of 3% real. This and other
26 peculiarities of the NYSEP are discussed in Attachment 3.

27 ⁹³NYSEO describes this value as being in nominal dollars, but
28 the derivation implies that it is in real 1990\$.

1 optimal emission level to be 30% below the allowance
2 level, with a marginal value of further reductions of
3 \$858/ton (\$1990). These valuations include only human
4 mortality, without any value for morbidity (illness),
5 visibility, acid rain, or other effects of SO₂. Hence,
6 these values are probably understated, and are certainly
7 additive with the acid-rain related allowance cost.

8 NO_x: NYSEO estimates a marginal cost of control of \$4,204/ton
9 (\$1990), up from \$2,461/ton NO_x in the 1989 Final Draft
10 Report. This value is based on the average cost of low-
11 NO_x burners (LNB) and SCR for NO_x control on coal-fired
12 power plants. NYSEO believes that SCR on coal plants
13 will be necessary to meet the air quality standard of
14 Title I of the CAAA. NYSEO estimates that the
15 incremental cost of SCR, on top of LNB, would be about
16 \$8,800/ton.⁹⁴

17
18 CO₂: NYSEO finds that stabilizing CO₂ emissions by 2008 will
19 require measures costing \$300/ton C (\$82/ton CO₂), while
20 reductions of more than 5% will require measures costing
21 \$500/ton C (\$136/ton CO₂). This cost is partially offset
22 by the SO₂ and NO_x benefits of the CO₂ control measures.
23 NYSEO estimates this offset at about \$16/ton on an
24 average-cost basis; on a marginal basis, the offset might
25 be worth as much as \$80/ton, bringing the net marginal

26 ⁹⁴See discussion in Attachment 3.

1 costs at stabilization and at reduction to \$220/ton C
2 (\$60/ton CO₂) and \$420/ton C (\$115/ton) CO₂.
3

4 The New York PSC has also ordered the utilities to fund
5 an environmental costing study of the direct costs of full
6 fuel cycle externalities of power generation in New York.
7 This study, cofunded by NYSERDA, ESEERCO, and EPRI will take
8 approximately four years and over one million dollars, and
9 will focus on developing direct costs of specific external
10 effects.
11

12 5.5 South Carolina

13 Q: How has the South Carolina Public Service Commission addressed
14 the issue of externalities?

15 A: The South Carolina Public Service Commission recently ordered
16 utilities to monetize environmental externalities of supply-
17 and demand-side options "where sufficient data is available,"
18 and evaluate externalities qualitatively where not enough
19 information exists on their costs.⁹⁵ The staff emphasized:
20

21 environmental costs should be monetized and
22 included within the planning process whenever
23 possible . . . each utility should identify
24 and monetize, to the extent possible, the cost
25 of compliance for existing and projected
26 supply-side options.⁹⁶

27 ⁹⁵South Carolina Public Service Commission Staff Docket 87-
28 223-E, Integrated Resource Planning Process. August 28, 1991.

29 ⁹⁶South Carolina Public Service Commission Staff Docket 87-
30 223-E, Integrated Resource Planning Process. August 28, 1991.

1 The PSC has not yet established a specific methodology
2 for utilities to use in monetization or evaluation. The PSC
3 also ordered the utilities to include the internalized costs
4 of complying with both current and anticipated environmental
5 regulations as part of the integrated resource planning
6 process.

8 5.6 Wisconsin

9 Q: Has Wisconsin updated its treatment of externalities?

10 A: Since 1989, the Wisconsin Public Service Commission has
11 discounted non-combustion energy options by 15 percent. In
12 its 1991 assessment of the state's electric utility plans,
13 however, the Commission staff noted that the 15% credit may
14 be too low. Among the advantages of monetization cited by the
15 staff was its ability to distinguish differences among
16 resources and make comparisons among options easier.⁹⁷

18 5.7 Bonneville Power Administration

19 Q: How does the Bonneville Power Administration address the issue
20 of externalities?

21 A: The Environmental Cost Work Group was formed in 1990 to
22 determine externality values for resource options in the
23 Northwest. Resources evaluated by the workgroup include
24 pulverized coal; atmospheric fluidized bed combustion (AFBC);

25 ⁹⁷Wisconsin Public Service Commission, Staff Assessment of
26 Electric Utility Plans: Advance Plan 6. July 1991.

1 integrated gasification combined cycle (IGCC); gas turbines;
2 cogeneration fired by natural gas, biomass, and municipal
3 solid waste; stand-alone municipal solid waste; new hydro and
4 additions to existing hydro; geothermal; solar; wind; and
5 conservation. Preliminary externality values were determined
6 for several externalities, among them air emissions SO_2 , NO_x ,
7 CO_2 , and total suspended particulates, and proxies for water
8 and land use. Those values are listed in Figure 5.7.
9 However, the Work Group recently dropped the CO_2 value from
10 its analysis, citing uncertainty in costs. The group has also
11 limited its fish impacts analysis in light of two other
12 parallel efforts on the same topic. Where possible, the Work
13 Group relied on direct environmental cost estimates developed
14 by ECO Northwest.⁹⁸ The externality values derived from these
15 studies are substantially understated, due to understatements
16 of the value of human life, and to the limitations of the
17 range of effects considered. The Work Group continues to meet
18 to discuss the issues.

21 ⁹⁸The ECO Northwest studies (1985, 1986, 1988) have been
22 previously reviewed in Chernick and Caverhill, 1989 and Ottinger,
23 et al. (1990).

1 6. CONCLUSIONS

2 Q: What are your conclusion and recommendations with respect to
3 the externalities that the Department has already valued?

4 A: The DPU should at least maintain its current estimates of the
5 values of externalities. Upward adjustments of externality
6 costs would be justified, up to about:

- 7 • \$4,500/T for SO₂, based on direct costing for health
8 effects;
- 9 • \$300/T for CO₂, based on estimated costs of control;
- 10 • \$8,800/T for NO_x, based on New York's estimate of
11 measures required for compliance with the Clean Air Act
12 Amendments ozone limits; and
- 13 • \$10,000/T for particulates.

14 Smaller increases may be justified by considerations of
15 continuity. Also, once the more modest externality values
16 already adopted by the DPU have been implemented by all
17 electric and gas utilities, and once other regulators adopt
18 similar approaches, the cost of reaching environmental targets
19 will decrease, and the marginal costs of compliance and the
20 marginal damage costs may also fall. No electric utility has
21 yet incorporated the DPU's externality values in an integrated
22 plan; until the DPU determines the effect of the established
23 values, it may not want to radically increase those values.

24 Modest increases, at least covering inflation, should be
25 applied to all the existing values, and those values should

1 be at least rounded up, to reflect the likely understatement
2 of costs.

3 Q: What are your conclusion and recommendations with respect to
4 the valuation of additional externalities?

5 A: We recommend that the Department adopt valuations for ozone
6 depletion, air toxics, thermal pollution, and oil imports.

7 For ozone depletion, we recommend a value of \$67,600/ton
8 (or \$34/lb) of CFC-11 equivalent eliminated. This is a value
9 per pound installed in equipment, only a small part of which
10 leaks out each year. The value of reducing leakage would be
11 higher.

12 For air toxics, we recommend initial values in Section
13 4.2. The most important value is that for residual plants,
14 for which we suggest 1¢/kWh. The actual costs of air toxics
15 may be much higher.

16 For thermal pollution, we recommend a value of 80¢/MMBTU
17 rejected to water, or about 0.4¢/kWh for typical existing
18 plants.

19 For oil imports, we recommend that the Department adopt
20 a value of \$2.50/MMBTU in 1991\$, based on the analysis we
21 presented in DPU 89-239.

22 Q: Does this conclude your testimony?

23 A: Yes.

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Table 3.1.1
Extrapolation of Particulate Matter Valuation to SO2

| | Ambient Air Quality Standard ($\mu\text{g}/\text{m}^3$) | Relative Hazard Per Pound | Value Per Lb If PM Is Worth \$2/Lb |
|-----|---|---------------------------------|--|
| | [1] | [2] | [3] |
| PM | 150 | 1.00 | \$2.00 |
| SO2 | 365 | 0.41 | \$0.82 |

Notes:

[1]: National Ambient Air Quality Standards or Massachusetts Ambient Air Quality Standards, in micrograms per cubic meter, 24-hour simple average.

[2]: $([1] \text{ for PM} / [1])$

[3]: $[2] * (\$2/\text{lb})$

Table 3.1.2: Boston Edison Edgar – Costs of Switching from .3% S oil to .2% S oil

| Boston Edison Forecast Levelized Fuel Costs (1990\$/MMBtu) | | #2 oil: | |
|--|---------|---------------------|--------------------------------|
| 0.3% S No. 2 Fuel Oil | \$10.97 | BTU/lb | 19,200 (Fink & Beatty, p. 5-5) |
| 0.2% S No. 2 Fuel Oil | \$11.03 | lb/MMBTU | 52 |
| Percentage Increase | 0.55% | sulfur differential | lb/MMBTU 0.052 |

| Year | Fuel price 0.3% S oil (Current \$/MMBtu) | GNP deflator | Fuel price 0.3% S oil (1990\$/MMBtu) | Fuel price 0.2% S oil (1990\$/MMBtu) | Price Differential (1990\$/MMBtu) | Cost of Sulfur Reduction | |
|------|--|--------------|--|--|---|--------------------------|-----------------------|
| [1] | [2] | [3] | [4] | [5] | [6] | (1990\$/lb S) [7] | (1990\$/ton S) [8] |
| 1990 | \$4.46 | 131.60 | \$4.46 | \$4.48 | \$0.0245 | \$0.471 | \$942 |
| 1991 | \$4.82 | 137.40 | \$4.62 | \$4.64 | \$0.0254 | \$0.488 | \$975 |
| 1992 | \$5.14 | 144.40 | \$4.68 | \$4.71 | \$0.0258 | \$0.495 | \$989 |
| 1993 | \$5.46 | 151.70 | \$4.74 | \$4.76 | \$0.0261 | \$0.500 | \$1,000 |
| 1994 | \$5.81 | 159.20 | \$4.80 | \$4.83 | \$0.0264 | \$0.507 | \$1,014 |
| 1995 | \$6.25 | 167.30 | \$4.92 | \$4.94 | \$0.0270 | \$0.519 | \$1,038 |
| 1996 | \$6.78 | 176.10 | \$5.07 | \$5.09 | \$0.0279 | \$0.535 | \$1,070 |
| 1997 | \$7.50 | 185.60 | \$5.32 | \$5.35 | \$0.0292 | \$0.562 | \$1,123 |
| 1998 | \$8.30 | 195.80 | \$5.58 | \$5.61 | \$0.0307 | \$0.589 | \$1,178 |
| 1999 | \$9.32 | 206.80 | \$5.93 | \$5.96 | \$0.0326 | \$0.626 | \$1,253 |
| 2000 | \$10.42 | 218.50 | \$6.28 | \$6.31 | \$0.0345 | \$0.663 | \$1,325 |
| 2001 | \$11.69 | 230.50 | \$6.67 | \$6.71 | \$0.0367 | \$0.705 | \$1,410 |
| 2002 | \$13.05 | 243.10 | \$7.06 | \$7.10 | \$0.0389 | \$0.746 | \$1,492 |
| 2003 | \$14.49 | 256.10 | \$7.45 | \$7.49 | \$0.0410 | \$0.786 | \$1,573 |
| 2004 | \$15.79 | 269.70 | \$7.70 | \$7.75 | \$0.0424 | \$0.814 | \$1,627 |
| 2005 | \$17.09 | 283.90 | \$7.92 | \$7.97 | \$0.0436 | \$0.837 | \$1,673 |
| 2006 | \$18.43 | 298.80 | \$8.12 | \$8.16 | \$0.0446 | \$0.857 | \$1,714 |
| 2007 | \$19.85 | 314.40 | \$8.31 | \$8.35 | \$0.0457 | \$0.877 | \$1,755 |
| 2008 | \$21.18 | 330.80 | \$8.43 | \$8.47 | \$0.0463 | \$0.890 | \$1,780 |
| 2009 | \$22.50 | 347.80 | \$8.51 | \$8.56 | \$0.0468 | \$0.899 | \$1,798 |
| 2010 | \$23.82 | 365.90 | \$8.57 | \$8.61 | \$0.0471 | \$0.905 | \$1,809 |
| 2011 | \$25.25 | 385.00 | \$8.63 | \$8.68 | \$0.0475 | \$0.911 | \$1,823 |
| 2012 | \$26.68 | 405.20 | \$8.67 | \$8.71 | \$0.0477 | \$0.915 | \$1,830 |
| 2013 | \$28.26 | 426.50 | \$8.72 | \$8.77 | \$0.0480 | \$0.921 | \$1,842 |
| 2014 | \$29.93 | 448.70 | \$8.78 | \$8.83 | \$0.0483 | \$0.927 | \$1,854 |

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[5] [4] * 1.0055

[6] [5] - [4]

[7] [6] / .052

[8] [7] * 2000

Table 3.1.3
Summary of New SO₂ Externality Value Estimates

| <u>Estimate Source</u> | <u>Value per Ton of Avoided Emissions in MA</u> | <u>Notes</u> |
|---|---|---|
| Mass. DPU PM value and DEP Ambient Air Quality Standards | \$1,640 | In 1989\$. |
| Draft NY State Energy Plan 1991 | \$4,500 | After meeting CAAA Title IV, adjusted for MA population density. |
| Switching Edgar from .2% to .3% sulfur oil | \$1,000– \$1,900 | Actual cost depends on year. All in 1990\$ |

Derivations can be found in text.

Figure 3.2.1

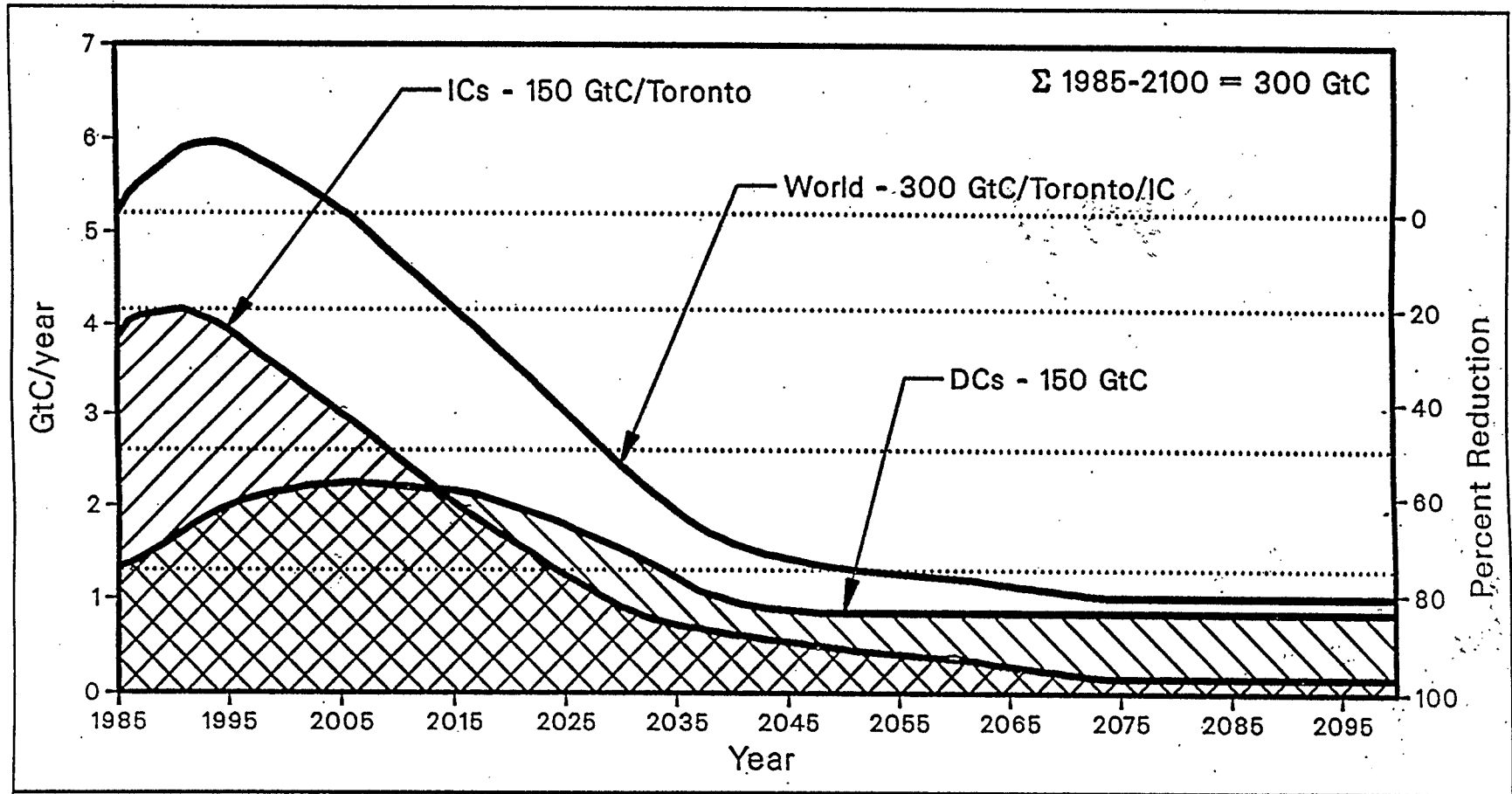
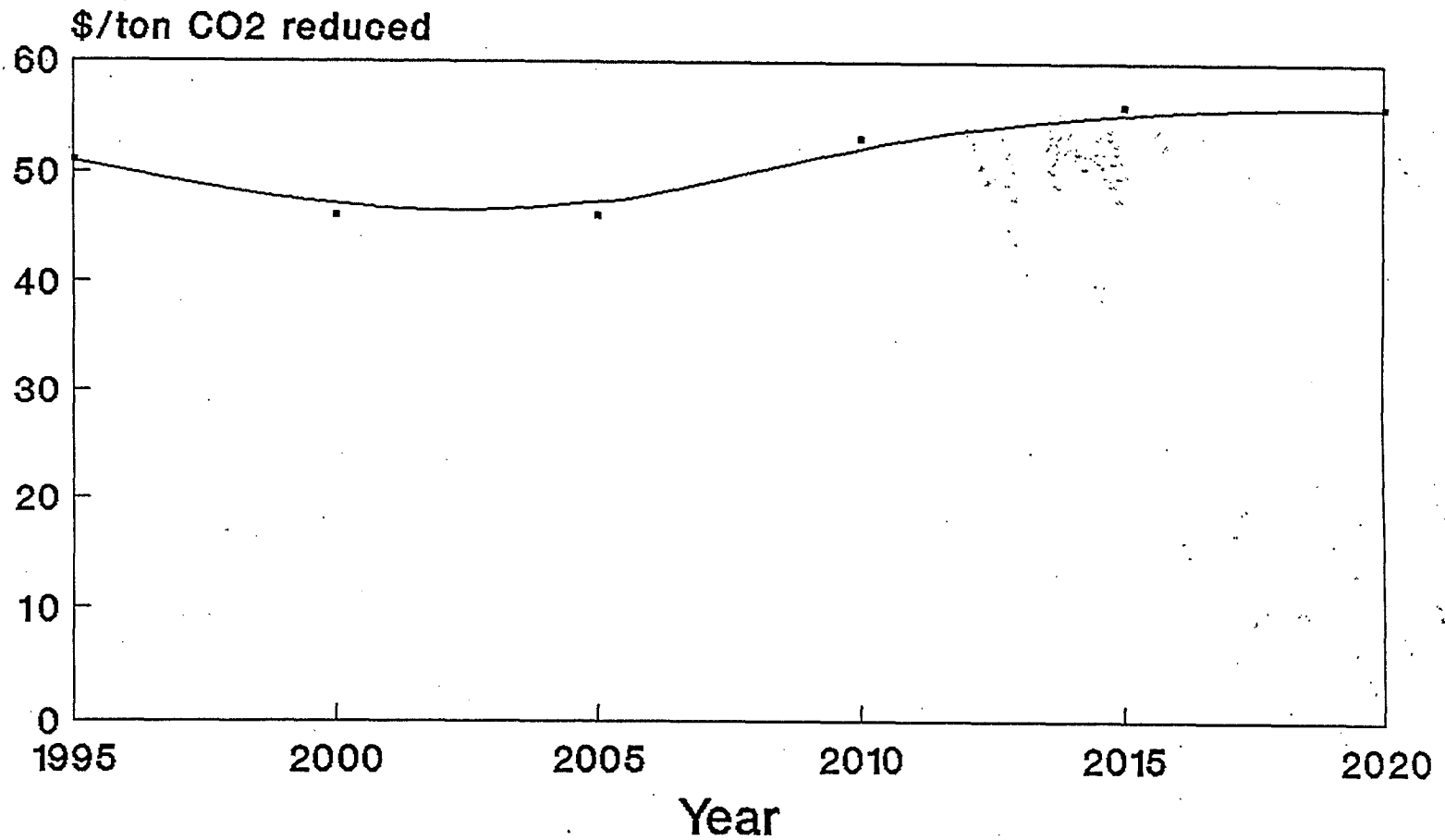


Figure I.6.2 Application of Toronto Target to Industrialized Countries

SOURCE: Krause, Bach, and Koomey. "Energy Policy in the Greenhouse," 1989.

Figure 3.2.2

Jorgenson & Wilcoxon
Marginal Cost of CO₂ Reduction



Source: Jorgenson, D.W. and Wilcoxon, P.J., "Reducing US Carbon Dioxide Emissions: The Cost of Different Goals"

Costs are in 1990\$

Table 3.2.1

Selected CO2 Reduction Targets

Sept 26, 1991

| Source | Target for CO2 Emission Reductions | Implied % reduction from base, assuming base annual growth of CO2 emissions of: | |
|------------------------|--|---|--------|
| | | 2% | 1% |
| [1] IPCC | Over 60% immediate reduction needed to stabilize concentrations at today's levels. | NA | NA |
| [2] Krause, et al. | 25% reduction required by industrialized countries from 1990 levels by 2005. | 44% | 35% |
| | 50% reduction required by industrialized countries from 1990 levels by 2015. | 70% | 61% |
| [3] Canada | Stabilization at 1990 levels by 2000. | 18% | 9% |
| [4] United Kingdom | Stabilization at 1990 levels by 2005. | 26% | 14% |
| [5] Norway | Stabilization at 1990 levels by 2000. | 18% | 9% |
| [6] Japan | Stabilization at 1990 levels by 2000. | 18% | 9% |
| [7] Sweden | Stabilization at 1990 levels by 2000. | 18% | 9% |
| [8] Denmark | 20% reduction from 1990 levels by 2000. | 34% | 27% |
| [9] Netherlands | 3-5% reduction from 1989-90 levels by 2000. | 20-22% | 12-14% |
| [10] Austria | 20% reduction from 1990 levels by 2005. | 41% | 31% |
| [11] New Zealand | 20% reduction from 1990 levels by 2000. | 34% | 27% |
| [12] Oregon | 20% reduction from 1990 levels by 2005. | 41% | 31% |
| [13] Germany | 25% reduction from 1990 levels by 2005. | 44% | 35% |
| [14] Toronto | 20% reduction from 1988 levels by 2005. | 43% | 32% |
| [15] Australia | Stabilization of 1988 levels by 2000. | 21% | 12% |
| | 20% reduction from 1988 levels by 2005. | 43% | 32% |
| [16] France | Stabilization at 1990 levels by 2005. | 26% | 14% |
| | 20% reduction from 1990 levels by 2025. | 60% | 44% |
| [17] Urban CO2 Project | 1-2% reduction per year. | NA | NA |

Sources:

- [1]: Global Environmental Change Report, Vol II, No. 11 (6/8/90). p. 4.
 [2]: Krause, Bach and Koomey, "Energy Policy in the Greenhouse," Vol 1 (1989), figure 1.6.2.
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 [11],[13]: Science News, Mar 1991.
 [14]: Global Environmental Change Report, Vol. III, No. 7, April 5, 1991.
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 [16]: Global Environmental Change Report, Vol. II, No. 19, October 12, 1990.
 [17]: The regions affiliated with the Urban CO2 Project are Toronto, Denver, Minneapolis/St. Paul, Portland (Ore.), Dade County (Fla.), San Jose (Calif.), Hannover (Germany), Saarbrucken (Germany), Copenhagen, Helsinki, and Ankara. From Global Environmental Change Report, Vol. III, No. 12, June 21, 1991.

Table 3.2.2

Estimates of the Cost of CO2 Emission Reductions.

Page 1 of 5.

| Source and Measure [a] | Cost of reduction (1990\$/T CO2) [b] | Percent reduction from base [c] |
|---|---|--|
| [1] U.S. EPA <u>CO2 scrubbing</u> | \$39 - \$51 | 90% of plant stack emissions controlled |
| [2] <u>Naill, Belanger and Petersen</u> Conservation | | |
| high | negative | 18% reduction from base |
| <u>very high</u> | \$69 | 28% reduction from base |
| Reforestation offsets | \$22 | 55% reduction from base |
| Coal efficiency tax | \$65 | 12% reduction from base |
| Carbon tax | | |
| <u>\$91/Ton C</u> | \$140 | 31% reduction from base |
| \$227/Ton C | \$176 | 51% reduction from base |
| \$364/Ton C | \$219 | 53% reduction from base |
| \$568/Ton C | \$273 | 57% reduction from base |
| [3] New York State Energy Office <u>CO2 scrubbing (coal plant)</u> | \$43 | reduction of 20% of 1988 levels by 2000. |
| [4] New York State Energy Plan (1989) <u>CO2 scrubbing (coal plant)</u> CO2 scrubbing (oil plant) | \$25 \$37 | reduction of 20% of 1988 levels by 2000. |
| [5] NYSEO (FRG externalities workshop) <u>utility sector mix (tree</u> planting, conservation, fuel switching, renewables, etc...) | \$48 \$91 \$136 \$167 | 31% reduction from base by 2008 36% reduction from base by 2008 39% reduction from base by 2008 43% reduction from base by 2008 |

cont...

Table 3.2.2 continued

Page 2 of 5.

| | Source and Measure [a] | Cost of reduction (1990\$/T CO ₂) [b] | Percent reduction from base [c] |
|------|--|---|--|
| [6] | Manne and Richels \$250/Ton carbon tax | -- | 20% reduction of 1990 emissions by 2020 and stabilization thereafter. |
| [7] | Steinberg and Cheng CO ₂ scrubbing (coal plant) | \$58 | 90% of plant stack emissions controlled |
| [8] | Nordhaus mix (sequestration, emission reduction) | \$23 \$28 \$48 \$78 \$119 | 17% from base emissions 21% from base emissions 25% from base emissions 34% from base emissions 42% from base emissions |
| [9] | Spectrum Economics utility sector mix (tree planting, conservation, fuel switching, renewables, etc...) | \$49 \$88 \$172 \$261 | 25% reduction from base by 2008 29% reduction from base by 2008 33% reduction from base by 2008 37% reduction from base by 2008 |
| [10] | Chernick and Caverhill Carbon sequestration (trees) | \$23 | N/A |
| [11] | DOE, Office of Energy Research fuel switching coal 1995 to gas 2010 | \$98 \$222 | N/A |
| [12] | Worldwatch Institute improving energy efficiency wind power geothermal power wood power steam inj. GT solar-thermal (gas) nuclear power photovoltaics CC coal | < 4.58 \$27 \$32 \$36 \$51 \$52 \$153 \$235 \$273 | N/A N/A |

Table 3.2.2, cont.

Estimates of the Cost of CO2 Emission Reductions.

Page 3 of 5.

| Source and Measure | | Cost of reduction (1990\$/T CO2) | Percent reduction from base |
|--------------------|---|--|-----------------------------------|
| [a] | | [b] | [c] |
| [13] | <u>World Wildlife Fund</u> | | |
| | U.S.A. | | |
| | Natural gas replacing coal | \$145 | 8% |
| | Gas combined cycles | \$21 | 11% |
| | Nuclear | \$12 | 14% |
| | Biomass as boiler fuel | \$54 | 14% |
| | Biomass liquid fuels | \$75 | 14% |
| | United Kingdom | | |
| | Nuclear/non-fossil | \$244 | NA |
| | Poland | | |
| | All energy conservation options | \$1 | 51% |
| | Marginal conservation options | \$7 | NA |
| | USSR | | |
| | Additional renewables | \$12 | NA |
| | CO2 scrubbers | \$18 | 1% |
| | Japan | | |
| | \$95/ton CO2 tax | \$6,096 | 18% |
| [14] | <u>Danish Ministry of Energy</u> | | |
| | Change from an economic growth scenario to an environmental growth scenario | | |
| | in 2000 | \$68 | 12% |
| | in 2015 | \$131 | 10% |
| | in 2030 | \$182 | 12% |
| | Heat conservation in existing buildings | \$107 | 11% |
| | Heat conservation in new buildings | \$41 | 2% |
| | More efficient electricity production | \$19 | 29% |
| | Renewable energy | \$6 | 26% |

Table 3.2.2, cont.

Estimates of the Cost of CO2 Emission Reductions.

Page 4 of 5.

| Source and Measure | | Cost of reduction (1990\$/T CO2) | Percent reduction from base |
|--------------------|--|--|-----------------------------------|
| | [a] | [b] | [c] |
| [15] | <u>Oregon Department of Energy</u> | | |
| | Convert public and private fleets to natural gas | \$296 | 0.02% |
| | Convert intra-city buses to natural gas | \$108 | 1% |
| | No new coal plants; Back down some coal plants after 1997; Build 900 MW renewable energy | \$43.5-\$56 | 5% |
| [16] | <u>Jorgenson & Wilcoxon</u> | | |
| | Switching from one CO2 emissions reduction target to a more strict target | \$56 | 15% |

- [b]: 4% annual inflation assumed.
- [1]: U.S. Environmental Protection Agency, "Policy Options for Stabilizing Global Climate," draft report to Congress (2/89) Vol II, p. V11-135. Assumes CO₂ emissions of 2 lb/kWh.
- [2]: Naill, Belanger, Petersen, "A Least-Cost Strategy for CO₂ Reduction," from NARUC National Conference on Environmental Externalities (10/90), Table 4.
- [3]: New York State Energy Office Division of Policy Analysis and Planning, "Environmental Externality Issue Report" (2/89), Preliminary Draft, p. 11.
- [4]: New York State Energy Office, NYS Dep't of Public Service, NYS Dep't of Environmental Conservation, "Draft New York State Energy Plan; Issue 2b: Air Impacts, Electricity," (5/89) p. 36. New York could meet its 20% goal through tree planting and coal plant scrubbing; the 20% goal would not necessitate the more expensive oil plant scrubbing.
- [5]: NYSEO paper prepared by A. Sanghi for Oct. 1990 conference.
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- [11]: U.S. DOE, Office of Energy Research, "A Preliminary Analysis of U.S. CO₂ Emissions Reduction Potential from Energy Conservation and the Substitution of Natural Gas for Coal in the Period to 2010. Feb. 1989.
- [12]: Worldwatch Institute, Lester R. Brown, et al. "State of the World 1990."
- [13]: Chandler, W., "Carbon Emissions Control Strategies." World Wildlife Fund, 1990. In the U.S.A., we have assumed nuclear power costs 1.3 cents/kWh more than coal. Poland's energy conservation options include space heating management, reduction of transmission and distribution losses, buildings insulation, automation and measurement, existing industrial equipment, railway electrification, coal quality improvement, shift to diesel engines in light trucks, and new industrial technology. The marginal measure is new industrial technology. We have used the following exchange rates: .5822 pounds/\$, 2933 zlotys/\$, 16.92 rubles/\$ (commercial exchange rate), and 137.65 yen/\$.
- [14]: Danish Ministry of Energy, "Energy 2000." April 1990. The environmental scenario emphasizes reducing energy consumption. The economic scenario assumes all cost-effective reduction options have been carried out by 2000. Costs for the individual measures are average measure costs. Exchange rate = 6.585 krone/\$.
- [15]: Oregon Department of Energy, "Oregon Fourth Biennial Energy Plan." 1990.
- [16]: Jorgenson, D. and Wilcoxon, P., "Reducing US Carbon Dioxide Emissions: The Cost of Different Goals." 1990. The less strict reduction target is immediate stabilization of carbon emissions at their 1990 levels. The more strict goal is 20% reduction of 1990 emissions levels by 2005.

FIGURE 3.4

TABLE 3.2-2 DIFFERENTIAL CAPITAL AND LEVELIZED ANNUAL COSTS FOR
0.012 LB/MBTU PARTICULATE REMOVAL SYSTEM*

| | 0.012 lb/MBtu Particulate Emission (\$1,000) |
|--------------------------------------|---|
| Capital Costs | |
| Fabric filter | 1,410 |
| Ductwork and ID fans | 0 |
| Waste handling | 0 |
| 1990 capital cost | 1,410 |
| Contingency | 140 |
| 1990 Direct capital cost | 1,550 |
| Escalation | 180 |
| 1994 Direct capital cost | 1,730 |
| Indirects | 270 |
| Interest during construction | 460 |
| 1995 Total differential capital cost | 2,460 |
| Levelized Annual Costs | |
| Operating Personnel | 0 |
| Maintenance | 550 |
| Energy | 30 |
| 1994 levelized annual operating cost | 580 |
| Fixed charges on capital | 400 |
| 1995 total levelized annual cost | 980 |
| Incremental Particulate Removal, tpy | 26.3 |
| Incremental Removal Cost, \$/ton | \$37,260 |

*Costs are for fabric filter particulate removal system installed downstream of two circulating fluidized bed boilers.

SOURCE: AES Harriman Cove Cogeneration Project Air Emission License Application to the Maine Department of Environmental Protection, May 1, 1991. Page 3-13.

TABLE 3.2.3: WORLD WILDLIFE FUND SURVEY OF CARBON DIOXIDE EMISSIONS REDUCTION COSTS

| Country | Proposed measures [a] | Tons/year C reduced (in 2005 unless otherwise indicated) [b] | Cost (1990\$) [c] | \$/ton C (1990\$) [d] | \$/ton CO2 (1990\$) [e] | Necessary to achieve 20% reduction by 2005? [f] |
|-------------------|------------------------------------|--|-------------------------|-----------------------------|-------------------------------|---|
| 1. United States | Natural gas replacing coal | 130,000,000 | \$76,000,000,000 | \$585 | \$145 | ? |
| | Gas combined cycles | 180,000,000 | \$15,600,000,000 | \$87 | \$21 | Yes |
| | Nuclear | 240,000,000 | \$11,900,000,000 | \$50 | \$12 | Yes |
| | Biomass as boiler fuel | 240,000,000 | \$52,000,000,000 | \$217 | \$54 | Yes, without |
| | Biomass liquid fuels | 240,000,000 | \$72,800,000,000 | \$303 | \$75 | new nuclear |
| 2. United Kingdom | Nuclear/Non-fossil | NA | NA | NA | \$244 | ? |
| 3. Poland | All energy conservation potentials | 35,000,000 | \$156,000,000 | \$4 | \$1 | Yes |
| | Marginal measure | 33,000,000 | \$960,960,000 | \$29 | \$7 | Yes |
| 4. USSR | Additional renewables | NA | NA | \$49 | \$12 | Yes |
| | CO2 scrubbers | 50,000,000 | \$1,001,000,000 | NA | \$18 | ? |

Notes

1a. Nuclear power is assumed to cost 1.3 cents/kWh more than coal.

1b, 1c. Tons of carbon and costs are projected for the year 2010.

3a. Poland's energy conservation options include space heating management, reduction of transmission and distribution losses, buildings insulation, automation and measurement, existing industrial equipment, railway electrification, coal quality improvement, shift to diesel engines in light trucks, and new industrial technology. The marginal measure is new industrial technology.

4b. Figure for CO2 scrubbers is in tons/year CO2.

Source

Chandler, W., "Carbon Emissions Control Strategies." World Wildlife Fund, 1990.

Exchange Rates:

.5822 pounds/\$

2933 zlotys/\$

16.92 rubles/\$ (commercial exchange rate)

TABLE 3.2.4: CARBON EMISSIONS REDUCTION COSTS IN DENMARK

| Year | Change in Scenario [1] | <u>Economic Scenario</u> | | <u>Environmental Scenario</u> | | <u>Cost of Incremental Reduction</u> | |
|------|------------------------------|--------------------------------------|-------------------------------------|--------------------------------------|-------------------------------------|--|--------------------|
| | | CO2 Reduction from 1988 (Tons) | Average Cost (DKK0.01/ kg) | CO2 Reduction from 1988 (Tons) | Average Cost (DKK0.01/ kg) | DKK0.01/ kg | 1990\$/ ton CO2 |
| | | [2] | [3] | [4] | [5] | [6] | [7] |
| 2000 | C to B | 11,000,000 | -20 | 16,800,000 | 4 | 50 | \$68 |
| 2015 | C to B | 24,500,000 | -9 | 29,300,000 | 8 | 95 | \$131 |
| 2030 | C to B | 25,000,000 | -9 | 31,200,000 | 19 | 132 | \$182 |

Measure [8]

| | | | | | | | |
|------|---|--|--|--|--|----|-------|
| 2030 | Heat conservation in existing buildings | | | | | 85 | \$107 |
| 2030 | Heat conservation in new buildings | | | | | 33 | \$41 |
| 2030 | More efficient electricity production | | | | | 15 | \$19 |
| 2030 | Renewable energy | | | | | 5 | \$6 |

Source:

Danish Ministry of Energy, "Energy 2000." April 1990.

Notes:

[1] Scenario B is the environmental scenario, which emphasizes reducing energy consumption.

Scenario C is the economic scenario, which assumes that all cost-effective reduction options have been carried out by 2000.

[2], [4] Table 5.13.

[3], [5] Table 5.14, assuming "central" prices and 7% real discount rate.

[6] $([4] * [5] - [2] * [3]) / ([4] - [2])$

[7] Assumes 6.585 krone per dollar.

[8] Table 4.18. Costs are average measure costs.

TABLE 3.2.5: NAILL ET AL. COSTS OF CO2 REDUCTION FROM TAXES

| Tax Type | Amount (1990\$/ton C) | Incremental Cost (1990\$/ton CO2) |
|-----------------|----------------------------------|--|
| Coal efficiency | NE | \$65 |
| Carbon | \$91 | \$140 |
| Carbon | \$227 | \$176 |
| Carbon | \$364 | \$219 |
| Carbon | \$568 | \$273 |

Source: Naill, R., Belanger, S., and Petersen, E., "A Least-Cost Strategy for CO2 Reduction,"
 Proceedings of the NARUC Conference on Environmental Externalities. October 1990.

Table 3.3.1: Exhaust Emissions Standards for Cars and Light-Duty Trucks

| Program/ Standard | Year/ Category | <u>in grams/mile through 50,000 miles</u> | | | | |
|----------------------|-------------------|---|---------------------------------|-----|-----|--------------|
| | | Non-methane hydrocarbons | Non-methane organic gases | CO | NOx | Formaldehyde |
| Clean Air Act | 1994-95 | 0.25 | | 3.4 | 0.4 | |
| Clean Air Act | 1996-2000 | | 0.125 | 3.4 | 0.4 | 0.015 |
| Clean Air Act | 2001 | | 0.075 | 3.4 | 0.2 | 0.015 |
| LEV | Standard | 0.25 | | 3.4 | 0.4 | |
| LEV | TLEV | | 0.125 | 3.4 | 0.4 | 0.015 |
| LEV | LEV | | 0.075 | 3.4 | 0.2 | 0.015 |
| LEV | ULEV | | 0.040 | 1.7 | 0.2 | 0.008 |
| LEV | ZEV | | NA | NA | NA | NA |

Sources:

Clean Air Act Amendments of 1990

Massachusetts Department of Environmental Protection, "Background

Documents for Proposed Amendments to 310 CMR 7.00, et seq." September 1991.

Notes:

LEV = low emission vehicle. It is both the name of the program and of a category of vehicle under the program.

TLEV = transitional low emission vehicle.

ULEV = ultra-low emission vehicle.

ZEV = zero emission vehicle.

Table 4.1.1 Relative Global Warming and Ozone Depletion Potentials of Selected Gases

| <u>Gas</u> | <u>Global Warming Potential</u> (CO ₂ = 1) | <u>Ozone Depletion Potential</u> (CFC-11 = 1) | <u>Value of Reducing Emissions</u> (1990\$/ton) |
|------------------|--|--|--|
| [1] | [2] | [3] | [4] |
| CO ₂ | 1 | NA | NA |
| CO | 2.2 | NA | NA |
| CH ₄ | 10 | NA | NA |
| N ₂ O | 180 | NA | NA |
| CFC-11 | 1,300 | 1.0 | 67,600 |
| CFC-12 | 3,700 | 0.93 | 62,578 |
| CFC-113 | 1,900 | 0.83 | 56,108 |
| CFC-114 | 1,690 | 0.8 | 54,080 |
| CFC-115 | 13,800 | 0.38 | 25,688 |
| HCFC-22 | 410 | 0.05 | 3,380 |
| HCFC-134a | 338 | 0 | 0 |
| HCFC-123 | 26 | 0.02 | 1,352 |
| HCFC-124 | NA | 0.019 | 1,294 |
| HCFC-141b | NA | 0.088 | 5,949 |
| HCFC-142b | NA | 0.054 | 3,650 |
| HFC-125 | NA | 0 | 0 |
| HFC-134a | 400 | 0 | 0 |
| HFC-143a | NA | 0 | 0 |
| HFC-152a | 46 | 0 | 0 |
| NH ₃ | NA | 0 | 0 |

Notes:

- [2]: CO₂, CO, CH₄, N₂O, HCFC-22, CFC-11, and CFC-12 from Lashof and Ahuja. CFC-114, HCFC-134a, and HCFC-123 from York International, normalized to the CO₂ unit on the basis of CFC-11. CFC-113, CFC-115, HFC-134a, and HFC-152a from Epstein & Manwell. (Note: Epstein, et al. cited a GWP of 6,400 for CFC-114.)
- [3]: CFC-11, CFC-114, HCFC-22, HCFC-134a, HCFC-123, and NH₃ from York International. Fisher, et al. describe similar results obtained by four atmospheric modelling groups: Atmospheric and Environmental Research, Inc., Du Pont Central Research, Lawrence Livermore National Laboratory, and the University of Oslo. CFC-12, CFC-113, CFC-115, HCFC-124, HCFC-141b, HCFC-142b, HFC-125, HFC-134a, HFC-143a, and HFC-152a are from Fisher, et al., and are based on the means of both the 1-D and 2-D models presented. Due to uncertainties of the models, only one digit is significant, though two were presented.
- [4]: [3]*(\$67,600/ton). See text for derivation. Note that both the ODPs and this unit cost are normalized to CFC-11.

Sources:

- * Epstein, Gary, and S. Manwell. "An assessment of the environmental trade-offs between CFC use and lower efficiency cooling with alternative refrigerants," in DSM and the Global Environment. Synergic Resources Corp. April 1991.
- * Fisher, DA, CH Hales, DL Filkin, MKW Ko, ND Sze, PS Connell, DJ Wuebbles, ISA Isaksen, & F Stordal. "Model calculations of the relative effects of CFCs and their replacements on stratospheric ozone." Nature 344, 508-512.
- * Lashof, Daniel A. and Dilip R. Ahuja. "Relative global warming potentials of greenhouse gas emissions." Submitted to Nature February 1990. Authors of Natural Resources Defense Council and Tara Energy Research Institute, respectively.
- * York International. CFC Update, presented at International District Heating and Cooling Association. 1990 5th Annual Cooling Conference.

Figure 4.1.1

TABLE J.1 Cost Impact of CFC Phaseout--United States

| CFC Policy Option | CFC Reduction (Mt/yr) | CO ₂ - Equivalent Reduction (Mt/yr) | Capital Cost (M\$/life) | Equipment Lifetime (years) | Annual Capital Cost (M\$/yr) | | | Operat- ing Cost (M\$/yr) | Total Cost (M\$/yr) | | | Abatement Cost (\$/t CFC) | | | Abatement Cost (\$/t CO ₂ Equivalent) | | |
|---|--------------------------|---|-------------------------------|----------------------------------|---------------------------------------|-----|-----|------------------------------------|---------------------------|------|------|---------------------------------|-------|-------|---|-------|-------|
| | | | | | 3% | 6% | 10% | | 3% | 6% | 10% | 3% | 6% | 10% | 3% | 6% | 10% |
| Cleaning and blowing agents, aerosols, refrigerants, not-in- kind substitutes | 0.086 | 302 | 172 | 10 | 20 | 23 | 28 | -17 | 3 | 6 | 11 | 35 | 70 | 128 | 0.01 | 0.02 | 0.04 |
| Conservation and recycle | 0.098 | 509 | 74 | 5 | 16 | 18 | 20 | 0 | 16 | 18 | 20 | 163 | 184 | 204 | 0.03 | 0.04 | 0.04 |
| Cleaning and blowing agents, aerosols, fluorocarbon substitutes | 0.074 | 248 | 0 | 10 | 0 | 0 | 0 | 167 | 167 | 167 | 167 | 2250 | 2250 | 2250 | 0.67 | 0.67 | 0.67 |
| Refrigerants, fluorocarbon substitutes | | | | | | | | | | | | | | | | | |
| Chillers | 0.023 | 88 | 2,500 | 30 | 128 | 182 | 265 | 78 | 206 | 260 | 343 | 8956 | 11304 | 14913 | 2.35 | 2.97 | 3.92 |
| Mobile air conditioning | 0.030 | 170 | 5,000 | 10 | 586 | 679 | 814 | 135 | 721 | 814 | 949 | 24033 | 27133 | 31633 | 4.25 | 4.80 | 5.60 |
| Appliance | 0.002 | 11 | 1,067 | 15 | 89 | 110 | 140 | 9 | 98 | 119 | 149 | 49000 | 59500 | 74500 | 8.67 | 10.53 | 13.11 |
| Other | 0.010 | 67 | 1,500 | 10 | 176 | 204 | 244 | 45 | 221 | 249 | 289 | 22100 | 24900 | 28900 | 3.32 | 3.74 | 4.34 |
| Appliance insulation, fluorocarbon substitutes | 0.007 | 14 | 3,733 | 15 | 313 | 384 | 491 | 16 | 329 | 400 | 507 | 47000 | 57143 | 72429 | 23.59 | 28.69 | 36.36 |
| TOTAL | 0.33 | 1409 | 14,046 | | | | | | 1761 | 2033 | 2435 | | | | | | |

NOTE: Mt = megaton = 1 million tons. Tons are metric.

Source: NAS, "Policy Implications of Greenhouse Warming, 1991.

Figure 4.1.2

TABLE J.2 Cost Impact of CFC Phaseout--Worldwide

| CFC Policy Option | CFC Reduction (Mt/yr) | CO ₂ -Equivalent Reduction (Mt/yr) | Capital Cost (M\$/life) | Equipment Lifetime (years) | Annual Capital Cost (M\$/yr) | | | Operating Cost (M\$/yr) | Total Cost (M\$/yr) | | | Abatement Cost (\$/t CFC) | | | Abatement Cost (\$/t CO ₂ Equivalent) | | |
|--|-----------------------|---|-------------------------|----------------------------|------------------------------|------|------|-------------------------|---------------------|------|------|---------------------------|-------|-------|--|-------|-------|
| | | | | | 3% | 6% | 10% | | 3% | 6% | 10% | 3% | 6% | 10% | 3% | 6% | 10% |
| Aerosols, refrigerants, not-in-kind substitutes | 0.12 | 492 | 25 | 10 | 3 | 3 | 4 | -161 | -158 | -157 | -157 | -1316 | -1312 | -1307 | -0.32 | -0.32 | -0.32 |
| Conservation and recycle | 0.27 | 1402 | 203 | 5 | 44 | 48 | 53 | 0 | 44 | 48 | 53 | 164 | 178 | 198 | 0.03 | 0.03 | 0.04 |
| Cleaning and blowing agents, refrigerants, not-in-kind substitutes | 0.20 | 701 | 400 | 10 | 47 | 54 | 65 | -40 | 7 | 14 | 25 | 35 | 72 | 126 | 0.01 | 0.02 | 0.04 |
| Cleaning and blowing agents, aerosols, fluorocarbon substitutes | 0.21 | 705 | 0 | 10 | 0 | 0 | 0 | 473 | 473 | 473 | 473 | 2250 | 2250 | 2250 | 0.67 | 0.67 | 0.67 |
| Refrigerants, fluorocarbon substitutes | | | | | | | | | | | | | | | | | |
| Chillers | 0.04 | 152 | 3,750 | 30 | 191 | 272 | 398 | 135 | 326 | 407 | 533 | 8158 | 10185 | 13320 | 2.14 | 2.68 | 3.49 |
| Mobile air conditioning | 0.08 | 452 | 10,000 | 10 | 1172 | 1359 | 1628 | 360 | 1532 | 1719 | 1988 | 19154 | 21484 | 24844 | 3.39 | 3.80 | 4.40 |
| Appliance | 0.013 | 73 | 7,800 | 15 | 653 | 803 | 1026 | 59 | 712 | 862 | 1084 | 54762 | 66277 | 83385 | 9.69 | 11.73 | 14.76 |
| Other | 0.03 | 200 | 3,500 | 10 | 410 | 476 | 570 | 135 | 545 | 611 | 705 | 18177 | 20350 | 23487 | 2.73 | 3.05 | 3.52 |
| Appliance insulation, fluorocarbon substitutes | 0.037 | 74 | 22,200 | 15 | 1860 | 2286 | 2919 | 83 | 1943 | 2369 | 3002 | 52514 | 64027 | 81132 | 26.36 | 32.14 | 40.65 |
| TOTAL | 1.0 | 4251 | 47,878 | | | | | | 5424 | 6346 | 7706 | | | | | | |

NOTE: Mt = megaton = 1 million tons. Tons are metric.

Source: NAS, "Policy Implications of Greenhouse Warming, 1991.

Table 4.2.1

Toxic Chemicals in Combustion Flue Gas

| | |
|-------------------------------------|---------------------------------------|
| Acetaldehyde | Formaldehyde |
| Antimony Compounds | Hexachlorobenzene |
| Arsenic Compounds | Hydrochloric Acid |
| Benzene | Hydrofluoric Acid (Hydrogen fluoride) |
| Beryllium Compounds | Lead Compounds |
| Biphenyl | Manganese Compounds |
| Bis (2-ethylhexyl) phthalate (DEHP) | Naphthalene |
| Cadmium Compounds | Nickel Compounds |
| Carbon Disulfide | Pentachlorophenol |
| Carbon Tetrachloride | Phenol |
| Carbonyl Sulfide | Phosphorous |
| Chlorine | Selenium Compounds |
| Chlorobenzene | 2,3,7,8-Tetrachlorodibenzo-p-dioxin |
| Chloroform | Tetrachloroethylene (Perchloroethane) |
| Chromium Compounds | Toluene |
| Cobalt Compounds | Trichloroethylene |
| Dibenzofuran | 2,4,5-Trichlorophenol |
| 1,4-Dichlorobenzene (p) | |

Source: Chow, Winston, et al. (1990). "Managing Air Toxics". Presented at the 83rd Annual Air & Waste Management Association Meeting.

Table 4.2.2
Relative Toxicity of Toxic Air Emissions

| <u>Toxic Emission</u> | <u>Mass. Standard</u> (ug/m3) | <u>Connecticut Standard</u> (ug/m3) | <u>Virginia Standard</u> (ug/m3) | <u>Relative Toxicity</u> (Based on Lead) |
|-----------------------|----------------------------------|--|-------------------------------------|---|
| [1] | [2] | [3] | [4] | [5] |
| Arsenic | | 0.05 | 3.3 | 60 |
| Beryllium | 0.001 | | | 140 |
| Cadmium | 0.003 | | | 47 |
| Chromium | 1.36 | | | 0.10 |
| Copper | 0.54 | | | 0.26 |
| HCl | 2.03 | | | 0.07 |
| Lead | 0.14 | 3.00 | 2.50 | 1.0 |
| Manganese | | 20.00 | | 0.15 |
| Mercury | 0.14 | | | 1.00 |
| Nickel | 0.27 | | | 0.52 |
| POM | | | 7.00 | 0.36 |
| Selenium | 0.54 | | | 0.26 |
| Vanadium | 0.27 | | | 0.52 |
| Formaldehyde | 0.33 | | | 0.42 |

Notes:

- [2]: Massachusetts Department of Environmental Protection, Office of Research and Standards, "Threshold Effects Exposure Limits (TEL)," March 1989. Concentrations are for a 24-hour average.
- [3]: NATICH Data Base Report on State, Local, and EPA Air Toxics Activities, 1989. PB90-131459. Concentrations are averaged over an eight hour period. Connecticut was chosen for its geographical and political proximity to Massachusetts.
- [4]: NATICH Data Base Report on State, Local, and EPA Air Toxics Activities, 1989. PB90-131459. Concentrations are averaged over a twenty-four hour period. Virginia was the only state listed as having a polycyclic organic matter standard.
- [5]: The ratio of the acceptable ambient concentration of the indicated element to that of lead. For any given emission, both figures are based on the standards of only one state.

Table 4.2.3
Arsenic Control Costs

| <u>Plant</u> | <u>Revised EPA Annualized Cost</u> (1000\$/yr) | <u>Revised EPA Baseline Emissions</u> (Mg/yr) | <u>Control Equipment Efficiency</u> | <u>Average Unit Control Cost</u> (\$/lb controlled) |
|----------------------|---|--|---|--|
| [1] | [2] | [3] | [4] | [5] |
| ASARCO-Hay | 798 | 5.4 | 96% | 70 |
| ASARCO-El Paso | 379 | 9.9 | 96% | 18 |
| Kennecott-Utah | 2,028 | 1.5 | 96% | 640 |
| Kennecott-Hayden | 2,140 | 6.5 | 96% | 156 |
| Kennecott-McGill | 2,200 | 10.1 | 96% | 103 |
| Phelps Dodge-Morenci | 3,430 | 1.9 | 96% | 855 |
| Total | 10,975 | 35.3 | 96% | 147 |

Notes:

[1],[2]: EPA. Inorganic Arsenic Emissions from Primary Copper Smelters and Arsenic Plants - Background Information for Promulgating Standards. EPA-450/3-83-010b. May 1986. Table 8-3.

[3]: EPA-450/3-83-010b, Table 4-1. 1 Mg = 1 tonne = 1.10 short tons.

[4]: EPA-450/3-83-010b, Section 1-6.0.

[5]: $[2] * 1000 / ([3] * 1.10 * 2000 * [4])$.

Table 4.2.4
Valuation of Lead Implied By Valuation of SO2 and PM

| <u>Emission</u> | <u>Maximum 24-hour Air-Borne Concentration (ug/m3)</u> | <u>Implied Ratio of Health Risks</u> | | <u>DPU Valuation (1989\$/lb)</u> | <u>Implied Value of Lead (1989\$/lb)</u> |
|-----------------|--|---|---|--|--|
| | | <u>Ratio of Lead to Substance</u> | <u>Ratio of Substance to Lead</u> | | |
| | [1] | [2] | [3] | [4] | [5] |
| Lead | 0.14 | 1 | 1 | --- | |
| SO2 | 150 | 0.000933 | 1071 | \$2.00 | \$2,143 |
| PM | 365 | 0.000384 | 2607 | \$0.75 | \$1,955 |

Notes:

- [1]: Massachusetts Ambient Air Quality Standards.
- [2]: ([1] for lead)/[1]
- [3]: [1]/([1] for lead)
- [4]: These values were originally established by the Massachusetts DPU in Docket 89-239.
- [5]: [3]*[4]

Table 4.2.5
Summary of Lead Emissions Valuations

| <u>Control Context</u> | | <u>Implied Marginal Cost of Control Per Pound of As or Cr</u> | <u>Implied Marginal Cost of Control Per Pound of Lead</u> |
|------------------------|--|---|---|
| A. | Arsenic from primary copper smelters | \$855 | \$14 |
| B. | Arsenic from glass manufacturers | \$300 | \$5 |
| C. | Lead from secondary lead smelters | | \$500 |
| D. | Chromium in cooling towers | \$1,330 | \$1,300 |
| E. | Lead in paint | | \$15,000 -\$25,000 |
| F. | Mass DPU SO2 and PM externality values | | \$2,000 |

Derivations and caveats can be found in the text.

Table 4.2.6
Valuation of Air Toxics Emissions of Selected Electric Utility Plants

| Toxic Emission | Relative Toxicity (lb Pb eq/lb) | ESP Coal | | Scrubbed Coal | | Residual Oil | | Distillate Oil | |
|---|------------------------------------|-------------------------|----------|-------------------------|----------|-------------------------|----------|-------------------------|----------|
| | | lb/10 ¹² btu | lb Pb eq | lb/10 ¹² btu | lb Pb eq | lb/10 ¹² btu | lb Pb eq | lb/10 ¹² btu | lb Pb eq |
| [1] | [2] | [3] | [4] | [5] | [6] | [7] | [8] | [9] | [10] |
| Arsenic | 60 | 40.1 | 2406.0 | 17.2 | 1032.0 | 19.0 | 1140.0 | 4.2 | 252.0 |
| Beryllium | 140 | 3.0 | 420.0 | 0.1 | 15.4 | 4.2 | 588.0 | 2.5 | 350.0 |
| Cadmium | 47 | 9.2 | 429.3 | 1.0 | 45.5 | 15.7 | 732.7 | 10.5 | 490.0 |
| Chromium | 0.10 | 401.5 | 41.3 | 115.5 | 11.9 | 21.0 | 2.2 | 48.0 | 4.9 |
| Copper | 0.26 | 194.0 | 50.3 | 24.0 | 6.2 | 280.0 | 72.6 | 280.0 | 72.6 |
| HCl | 0.07 | 63,040.0 | 4347.6 | 3,940.0 | 271.7 | | | | |
| Lead | 1 | 49.0 | 49.0 | 16.8 | 16.8 | 28.0 | 28.0 | 8.9 | 8.9 |
| Manganese | 0.15 | 642.0 | 96.3 | 36.0 | 5.4 | 26.0 | 3.9 | 14.0 | 2.1 |
| Mercury | 1 | 12.0 | 12.0 | 4.2 | 4.2 | 3.2 | 3.2 | 3.0 | 3.0 |
| Nickel | 0.52 | 316.0 | 163.9 | 41.5 | 21.5 | 1260.0 | 653.3 | 170.0 | 88.1 |
| POM | 0.36 | 3.9 | 1.4 | 8.6 | 3.1 | 8.4 | 3.0 | 22.5 | 8.0 |
| Selenium | 0.3 | 1.6 | 0.4 | | | | | | |
| Vanadium | 0.52 | | | | | | | | |
| Formaldehyde | 0.42 | 9.3 | 4.0 | 8.6 | 3.6 | 0.02 | 0.01 | | |
| Totals (lbs OR lb Pb equiv.): | | 64721.6 | 8021.5 | 4213.4 | 1437.3 | 405.0 | 171.8 | 405.0 | 171.8 |
| Value @ \$150/lb Pb equiv. (\$/mmbtu): | | | \$1.20 | | \$0.22 | 2070.5 | 3398.7 | 968.6 | 1451.5 |
| Value @ \$1,500/lb Pb equiv. (\$/mmbtu): | | | \$12.03 | | \$2.16 | | \$0.51 | | \$0.22 |
| Value @ \$15,000/lb Pb equiv. (\$/mmbtu): | | | \$120.32 | | \$21.56 | | \$5.10 | | \$2.18 |
| | | | | | | | \$50.98 | | \$21.77 |

NOTES:

- [2]: Relative toxicity is the ratio of the acceptable ambient concentrations for each emission to that of lead. Whenever possible, relative toxicity was determined using Massachusetts standards. Arsenic and manganese were based on the ratios of the standards in Connecticut. POM was based on Virginia standards. Sources: Massachusetts Department of Environmental Protection, March 1989 standards. NATICH Data Base Report on State, Local, and EPA Air Toxics Activities, 1989. PB90-131459.
- [3],[5],[7],[9]: EPA-450/2-89-001: all emissions figures except for: selenium and vanadium, which come from DOE Technology Characterizations Handbook, 1981; and HCl, from T.E. Emmel, et al. (1989), Acidic Emissions Control Technology and Costs. Noyes Data Corp. ESP control for HCl from Winston Chow, et al. (1990) "Managing Air Toxics," presented at the 83rd Annual Air & Waste Management Meeting. EPRI 90.108.1.
- [4],[6],[8],[10]: [2]*(lb*10¹² for each source and each emission).

Table 4.2.7

Valuation of Toxic Air Emissions of Residual Oil-Fired Boilers with Electrostatic Precipitators

| Toxic Emission | Relative Toxicity (lb Pb eq/lb) | Uncontrolled Coal lb/10 ¹² btu | Coal With ESP lb/10 ¹² btu | Coal ESP Efficiency | Unscrubbed Residual Oil lb/10 ¹² btu | Residual Oil With ESP lb/10 ¹² btu lb Pb equiv | |
|---|------------------------------------|--|--|---------------------|--|---|--------|
| [1] | [2] | [3] | [4] | [5] | [6] | [7] | [8] |
| Arsenic | 60 | 684 | 40.1 | 94% | 19 | 1.1 | 66.8 |
| Beryllium | 140 | 81 | 3 | 96% | 4.2 | 0.2 | 21.8 |
| Cadmium | 47 | 44.4 | 9.2 | 79% | 15.7 | 3.3 | 151.8 |
| Chromium | 0.10 | 1410 | 401.5 | 72% | 21 | 6.0 | 0.6 |
| Copper | 0.26 | 848 | 194 | 77% | 280 | 64.1 | 16.6 |
| HCl | 0.07 | 78800 | | 20% | | | |
| Lead | 1 | 316 | 49 | 84% | 28 | 4.3 | 4.3 |
| Manganese | 0.15 | 2980 | 642 | 78% | 26 | 5.6 | 0.8 |
| Mercury | 1 | 16 | 12 | 25% | 3.2 | 2.4 | 2.4 |
| Nickel | 0.52 | 1160 | 316 | 73% | 1260 | 343.2 | 178.0 |
| POM | 0.36 | | 3.9 | | 8.4 | 8.4 | 3.0 |
| Selenium | 0.26 | 5 | 1.6 | 68% | | | |
| Vanadium | 0.52 | | | | 0.015 | | |
| Formaldehyde | 0.42 | | 9.315 | | 405 | 405.0 | 171.8 |
| Totals (lbs OR lb Pb equiv.): | | | | | | 843.6 | 618.0 |
| Value @ \$150/lb Pb equiv. (\$/mmbtu): | | | | | | | \$0.09 |
| Value @ \$1,500/lb Pb equiv. (\$/mmbtu): | | | | | | | \$0.93 |
| Value @ \$15,000/lb Pb equiv. (\$/mmbtu): | | | | | | | \$9.27 |

NOTES:

- [2]: Relative toxicity is the ratio of the acceptable ambient concentrations for each emission to that of lead. Whenever possible, relative toxicity was determined using Massachusetts standards. Arsenic, manganese and mercury were based on the ratios of the standards in Connecticut. POM was based on Virginia standards. Source: NATICH Data Base Report on State, Local, and EPA Air Toxics Activities, 1989. PB90-131459.
- [3],[4],[6]: EPA-450/2-89-001: all emissions figures except for: selenium and vanadium, which come from DOE Technology Characterizations Handbook, 1981; and HCl, from T.E. Emmel, et al. (1989), Acidic Emissions Control Technology and Costs. Noyes Data Corp. ESP control for HCl from Winston Chow, et al. (1990) "Managing Air Toxics," presented at the 83rd Annual Air & Waste Management Meeting. EPRI 90.108.1.
- [5]: $1 - [4]/[5]$
- [7]: $[6] * (1 - [5])$
- [8]: $[2] * [7]$

Table 4.3.1
Incremental Costs of Cooling Towers

| | | Capital or Present Value (\$M) | | Annual Cost (\$M) | | | Capacity | Heat Rejection | Control Cost |
|--------------------------|---------------------|-----------------------------------|----------------|-------------------|----------------|--------------|-------------|-------------------|-------------------|
| | | <u>Capital</u> | <u>O&M</u> | <u>Capital</u> | <u>O&M</u> | <u>Total</u> | <u>(MW)</u> | <u>(MMBtu/hr)</u> | <u>(\$/MMBtu)</u> |
| | | [1] | [2] | [3] | [4] | [5] | [6] | [7] | [8] |
| <u>Edgar</u> | | | | | | | | | |
| A. | Fan Wet Tower | 2.18 | 18.17 | 0.26 | 1.45 | 1.72 | 306 | 676 | 0.34 |
| B. | Natural Dry Tower | 7.48 | 28.68 | 0.90 | 2.29 | 3.19 | 306 | 676 | 0.63 |
| C. | Ironstone Wet Tower | 2.89 | 15.62 | 0.35 | 1.25 | 1.60 | 306 | 676 | 0.32 |
| <u>Manchester Street</u> | | | | | | | | | |
| D. | Wet Tower | 13.93 | | 1.67 | 4.30 | 5.97 | 457.5 | 1010.7 | 0.79 |

Notes:

- [1],[2] A,B: from Edgar Supplemental Draft EIR, page WQ-2-5.
C: from Edgar Long-Range Forecast. *Energy Park Project 5/1/90 EPSC*
D: from Environmental Assessment.
- [3]: [1]*12% real carrying charge
- [4]: A,B,C: [2]*8% real levelization
D: from Environmental Assessment.
\$4.3M/yr = 9 MW (penalty) * 85% (capacity factor) * 8760 hrs * 6.4 cents/kWh (escalated to 1990\$)
- [5]: [3]+[4]
- [6]: A,B,C: from Edgar Long-Range Forecast, page 2-3.
D: from Environmental Assessment, page 3-93.
- [7]: A,B,C: from Edgar Supplemental Draft EIR, page WQ-2-1.
D: [6] for D, times the ratio of [7] to [6] for A.
- [8]: [5]/([7]*8760*.85)*10⁶, assuming an 85% capacity factor.

Sources:

Edgar Supplemental Draft EIR:

United Engineers and Constructors, March 1990. Edgar Energy Park Supplemental Draft Environmental Impact Report.

Edgar Long-Range Forecast:

Boston Edison Company, May 1, 1990, before the Massachusetts Energy Facilities Siting Council.
Long-Range Forecast of Electric Power Needs and Requirements Edgar Energy Park Project.

Environmental Assessment:

Narragansett Electric Co. and New England Power Co., September 1989. Manchester Street Station Repowering Project Environmental Assessment.

4. DIRECT ASSESSMENT OF AN OIL IMPORT PREMIUM

About one-third of the oil consumed in the U.S. annually, or approximately 6 million bbls/day of crude and petroleum products, is imported.

The literature on the national costs of imported oil dates from the late 1980s. So far as we are aware, the most recent estimate of the national economic cost of oil imports (including vulnerability to interruptions and price swings, increases in inflation, deterioration of the balance of payments, and encouragement of further increases in oil prices) is in Broadman and Hogan (1988). This study estimates an expected oil import premium (taking into account the uncertainty in a range of parameters) of \$11.09/bbl in 1985\$. Assuming 5.7 MMBTU/bbl of crude oil, and 5% loss of energy content in the refining process, this is equivalent to \$2.05/MMBTU of oil products (including both #6 and #2 oil) in 1985\$, or \$2.26/MMBTU in 1988\$.

For comparison, Broadman (1986) reports a range of import premium estimates of \$2-\$124/bbl. CRA (1984) surveyed 17 estimates done between 1978 and 1981 and reported a similar range of results. Some estimates from the early 1980s were much higher than Broadman and Hogan (1988). Other estimates include only a subset of identified costs, and are therefore clearly understated. Recent estimates tend to cluster around the results of Broadman and Hogan (1988).

These estimates of oil import premiums include only effects on the United States. If the benefits to oil exporters of increased oil use in the U.S. were included in the analysis, the premium would be smaller. We would expect that policy makers and the public would generally be concerned about effects on the local population, which in the case of oil imports is the entire country. In addition, they may be concerned about those similarly situated (for oil imports, most of Western Europe, and Japan) or less advantageously situated (e.g., most of the Third World). However, some value might be assigned to the benefits of high oil prices to exporters who are otherwise similarly situated to the U.S. -- such as Canada, the U.K., or Norway -- or those which are disadvantaged in other respects, such as Mexico, Venezuela, or Nigeria.

Other international economic and political objectives may also affect the value of the oil import premium.

Table 5.1: Externality Values

| Externality | Mass DPU values (\$/lb) | Calif. PUC values (\$/lb) | Nevada PSC values (\$/lb) | Pace University values (\$/lb) | BPA East values (\$/lb) | BPA West values (\$/lb) | BPA values adjusted for Northeast (\$/lb) |
|---------------------------|----------------------------------|------------------------------------|------------------------------------|---|----------------------------------|----------------------------------|--|
| SO2 | 0.78 | 6.48 | 0.78 | 2.03 | 0.20 | 1.80 | 3.6-18 |
| NOx | 3.38 | 6.53 | 3.40 | 0.82 | 0.03 | 0.44 | .88-4.4 |
| CO | 0.45 | NE | 0.46 | NE | NE | NE | NE |
| PM10 (TSP for BPA values) | 2.08 | 4.39 | 2.09 | 1.19 | 0.08 | 0.77 | 1.54-7.70 |
| VOC's | 2.76 | 1.83 | 0.59 | NE | NE | NE | NE |
| CO2 | 0.011 | 0.004 | 0.011 | 0.0068 | 0.003 | 0.003 | 0.003 |

Sources:

California Public Utility Commission Decision 91-06-022. June 1991.

Massachusetts DPU Decision In Docket 89-239. August 31, 1990.

Nevada PSC Docket No. 89-752. January 22, 1991.

Ottinger, R. et al., "Environmental Costs of Electricity." Oceana; Dobbs Ferry, NY: 1990.

Bonneville Power Administration, "Environmental Costs and Benefits: Documentation and Supplementary Information." February 22, 1991.

Notes:

All values expressed in 1990\$.

The "BPA values adjusted for Northeast" are the BPA West values multiplied by 2-10 times to reflect the greater population density in the Northeast.

FIGURE 5.7

Bonneville Power Administration
DRAFT ENVIRONMENTAL COST ADJUSTMENTS
COMPETITIVE ACQUISITION OF FIRM ENERGY

February 22, 1991

(1990 mills/kWh)

| Resource Type | East | West |
|--|-------------------|------|
| Pulverized Coal | 9.6 | 15.5 |
| AFBC Coal (Fluidized Bed) | 8.5 | 10.4 |
| IGCC Coal (Coal Gasification) | 8.0 | 9.1 |
| Simple Cycle Combustion Turbine | 5.3 | 6.1 |
| Combined Cycle Combustion Turbine | 3.8 | 4.4 |
| New Hydro Facility | 2.0 | 2.0 |
| Natural Gas-Fired Cogeneration | 1.9 | 2.2 |
| Additions to Existing Hydro Facility | 1.0 | 1.0 |
| Geothermal | 0.1 | 0.1 |
| Wind | 0.1 | 0.1 |
| Solar | 0.1 | 0.1 |
| Conservation | 0 | 0 |
| Nuclear | Under development | |
| Wood-Fired Cogeneration | Under development | |
| Municipal Solid Waste-Fired Cogeneration | Under development | |

- These adjustments are subject to change based on an on-going review and are in real levelized 1990 mills/kWh. The adjustments will be finalized on April 15, 1991.
- Explanatory documentation of these adjustments can be obtained by calling BPA's document request line. Call 1-800-841-5867 (Oregon), 1-800-624-9495 (other western states), and 503-230-7334 (elsewhere).

(31421)

SOURCE: Bonneville Power Administration, February 22, 1991.
"Environmental Costs and Benefits: Documentation and
Supplementary Information."

Bonneville Power Administration

DRAFT ENVIRONMENTAL COST ADJUSTMENTS

COMPETITIVE ACQUISITION OF FIRM ENERGY

February 22, 1991

(1990 mills/kWh)

| Resource Type | East | West |
|--|-------------------|------|
| Pulverized Coal | 9.6 | 15.5 |
| AFBC Coal (Fluidized Bed) | 8.5 | 10.4 |
| IGCC Coal (Coal Gasification) | 8.0 | 9.1 |
| Simple Cycle Combustion Turbine | 5.3 | 6.1 |
| Combined Cycle Combustion Turbine | 3.8 | 4.4 |
| New Hydro Facility | 2.0 | 2.0 |
| Natural Gas-Fired Cogeneration | 1.9 | 2.2 |
| Additions to Existing Hydro Facility | 1.0 | 1.0 |
| Geothermal | 0.1 | 0.1 |
| Wind | 0.1 | 0.1 |
| Solar | 0.1 | 0.1 |
| Conservation | 0 | 0 |
| Nuclear | Under development | |
| Wood-Fired Cogeneration | Under development | |
| Municipal Solid Waste-Fired Cogeneration | Under development | |

- These adjustments are subject to change based on an on-going review and are in real levelized 1990 mills/kWh. The adjustments will be finalized on April 15, 1991.
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(31421)

SOURCE: Bonneville Power Administration, February 22, 1991.
 "Environmental Costs and Benefits: Documentation and
 Supplementary Information."

ATTACHMENT 3

REVIEW OF THE NEW YORK STATE ENERGY OFFICE

1991 EXTERNALITY ESTIMATES

In connection with the 1991 State Energy Plan, the New York State Energy Office (NYSEO) has updated its 1989 estimates of externality values. This information can be found in Issue Report 9 of the SEP, and in the **Analysis of Carbon Reduction in New York State**, June 1991. This review will discuss the derivation of these values, and their potential shortcomings, in terms of generic issues, SO₂ issues, NO_x issues, and CO₂ issues.

1. Generic Issues

a. Nominal and real costs

The 1991 Report characterizes the 1989 externality values, which the PSC has used as real-levelized values, as if they were nominally levelized values. However, the SEO's own 1989 reports state these costs in 1989\$.

For 1991, NYSEO presents undocumented cost estimates for SO₂ and NO_x, asserts that those values are nominally levelized, and then adjusts those values down to real-levelize them. NYSEO appears to be taking capital costs and present values, and levelizing them with a low discount rate or cost of capital. Personal communications with NYSEO indicate that the 1991 estimates use a 7% rate. This is too low even for a real-levelized analysis of capital costs; for a 30-year life, utility real-levelized carrying charges are typically in the 10.5% to 12% range. For the shorter lives SEO assumes for some retrofits, carrying charges should be even higher. The 7% value is roughly correct as a real utility discount rate. Hence, the NYSEO costs appear to be closer to real-levelized cost than to nominally-levelized costs.

b. Taxes

NYSEO appears to assume that externality costs can be internalized through "Trust Fund" (TF) taxes on emissions, where the tax is set at

$$C/T,$$

where

C is the cost of reducing the emissions by the desired amount (D), and

E is the total current emissions.

This value is lower than either the average cost of control (C/D), or the marginal cost of control (dC/dD). While this approach could theoretically offset emissions, it does not give polluters adequate incentives to reduce usage, since they will pay less for emitting than for controlling. Hence, it may be difficult to define the set of efficient controls, especially if the polluters are concerned that they may be required to undertake some of the controls without compensation from the trust fund.

2. SO₂ Issues

NYSEO estimates the marginal damage cost of SO₂, at an emission level equal to NY allowances under the 1990 Clean Air Act Amendments (CAAA), 280,000 Tons/yr, as \$2200/T in 1990\$. This damage cost is understated, since it includes only mortality, without morbidity, visibility, or other effects.

NYSEO then assumes that the marginal damage cost falls linearly to 0 as emissions fall to 100,000 tons/year. This is a very strong assumption, and is not supported by any data. NYSEO claims that "economic theory suggests" this shape for the marginal damage cost curve, but of course economic theory is irrelevant to estimating a dose-response curve.

If NYSEO were correct that marginal damage costs are linear, then damage cost at current emissions (380,000 tons) would be \$3400/T, falling gradually to \$2200 as the CAAA SO₂ provisions take effect by 2000.

NYSEO estimates a marginal cost of abatement, which starts with the cost of switching oil plants to lower-sulfur fuel. NYSEO assumes that 70,000 of the 100,000 T/yr of emission reductions required to live within the state's CAAA allowance limits are achieved by some other means, but does not specify them. These measures may include LILCo's reduction in oil sulfur content, use of summer gas, and installation of scrubbers on units targeted by the CAAA for 1995 reductions.

The marginal cost of control curve for remaining measures appears to be consistent with other estimates of the costs of fuel-switching and of scrubbing smaller, older, less utilized units. However, each increment of the curve lumps together a group of options. For example, NYSEO's option C is fuel-switching oil plants to 0.3% S oil. This actually includes relatively inexpensive reductions (e.g., 1.0% to 0.75%), and some very expensive reductions (e.g., 0.37% to 0.3%). Similarly, the scrubbing option will have a range of costs at different units.

NYSEO then plots its marginal damage curve against its marginal abatement-cost curve, and determines that a reduction of

75,000 T/yr beyond the CAAA requirements would be optimal. The intersection it finds, about \$1300/T, would probably be higher with a realistic cost curve.

NYSEO describes the resulting value as nominally-levelized, but the damage curve is definitely real-levelized and the cost curve also appear to be real-levelized.

The NYSEO results should be interpreted to indicate that the mortality value of SO₂ reductions are near \$3400/T now, falling to \$2200/T in 2000, and then falling further to something above \$1300/T whenever the additional reductions are complete. With acid rain, morbidity, visibility, and other effect, the value may be considerably higher.

3. NO_x Issues

NYSEO estimates that the marginal cost of required NO_x control (to meet Title I of the CAAA) is the installation of SCR (or the equivalent) on existing coal-fired plants, at a cost of \$6,100/T. However, of the 85% reduction achieved at this average cost, 50% is due to low-NO_x burners (LNB), which cost almost nothing in \$/T NO_x. The 1989 plan estimated a cost of \$395/T for LNB, and \$7,281 for SCR. Assuming \$870/T for LNB, which may be implied in Figure IV.7 of the Carbon Analysis, the marginal cost of adding SCR to LNB is \$13,600/T. Even assuming NYSEO is correct in describing this cost as nominally-levelized, the real-levelized cost would be about \$9400/T.

4. CO₂ Issues

NYSEO constructs a supply curve for CO₂ reductions, including a broad range of control options. One large set of options is lumped together as a "Low Emission Scenario;" it is not clear how much the individual options in this group cost. NYSEO finds that stabilization of emissions at 1988 levels would require measures costing up to \$300/T carbon, and that reductions of more than about 5% would require measures costing \$500/T carbon.

NYSEO credits these costs with about \$51/T in SO₂ and NO_x reductions, using the values from the Pace study. The value used for NO_x is only \$1640/T, which is clearly too low. The value of \$4060/T for SO₂ exceeds NYSEO's own estimate. Insufficient data are provided to allow for correction of these figures. In any case, this adjustment leaves NYSEO with a net carbon cost of \$240 to \$450/T, depending on the required reduction. (NYSEO's computations are not easy to follow at this point.)

NYSEO then converts these marginal costs to average costs, including measures with negative net carbon costs, which should be

pursued in any case. NYSEO's reported carbon "externality" value of \$8-\$50/T carbon is not an externality value at all, but a cost estimate for a hypothetical mitigation program. This value is useless in screening resources; indeed, using the NYSEO carbon "externality" values, most of the measures NYSEO has identified as necessary would be screened out.

The appropriate value for carbon externalities from the NYSEO 1991 studies would be \$240-\$450/T C, or \$66-125/T CO₂ (1990\$). This range is far higher than the values adopted by regulators to date.